

## Excess of Power During Electrochemical Loading: Materials, Electrochemical Conditions and Techniques

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## Remarks

Signals well above the measurement uncertainties are confirming the anomalous production of excess of power during electrochemical loading of Palladium with Deuterium.

Excess of power has the following features:

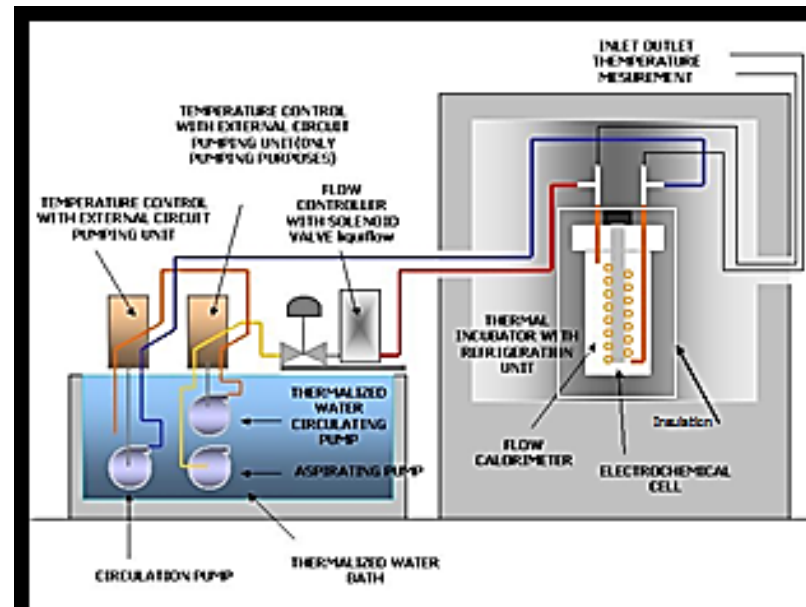
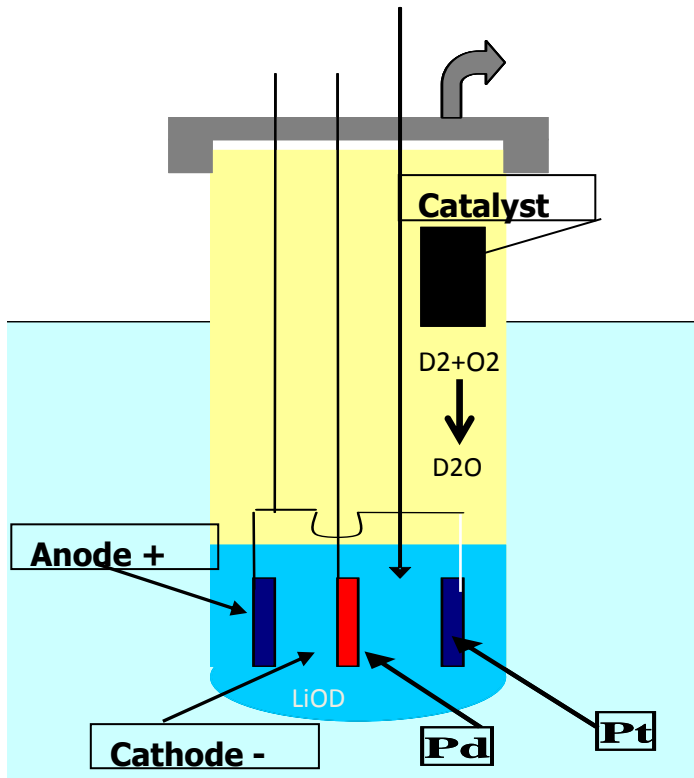
- 1) Threshold effect (loading  $D/Pd > 0.9$ )
- 2) Unobserved with hydrogen
- 3) Unexplainable as chemical effect
- 4) Occurs only if materials are showing specific characteristics

## Research Approach



- 1) Material science to increase both reproducibility and signals by loading enhancement.
- 2) Calorimetric experiments designed to have an appropriate signal/noise ratio.
- 3) Definition of the effect through the material characteristics.

## Calorimetry (Mass Flow): Closed Cells



Mass Flow Calorimeter

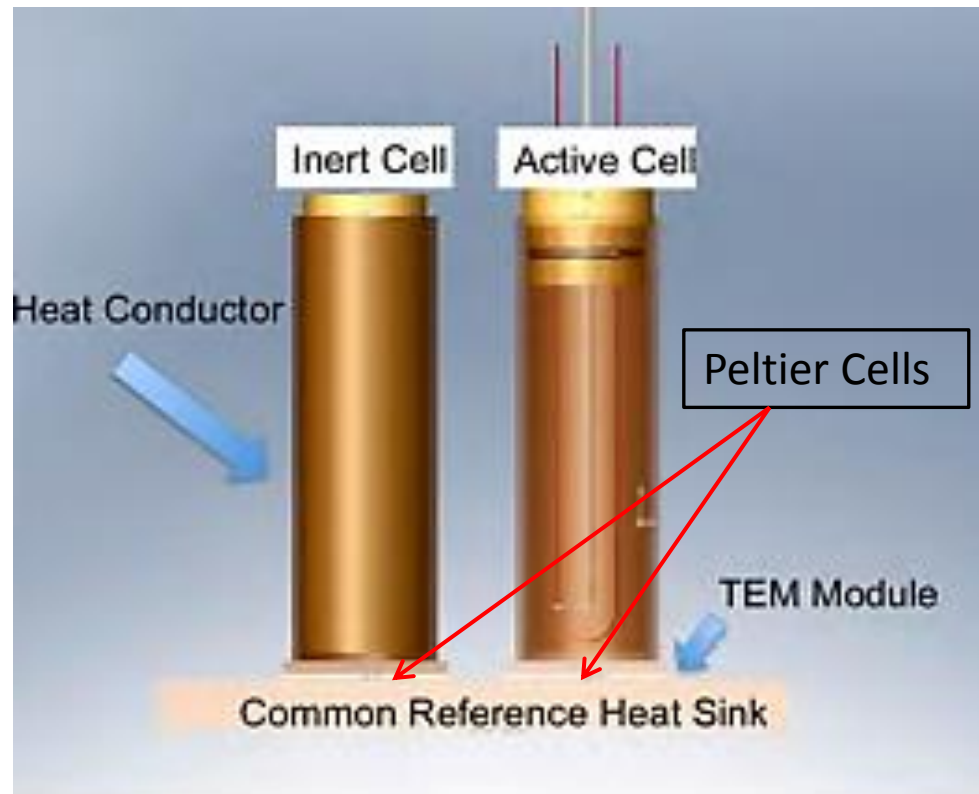
$$P_{IN} = VxI$$

$$P_{OUT} = WCp(T_{OUT} - T_{IN})$$

$$P_{EX} = P_{OUT}/\text{Efficiency} - P_{IN}$$

Closed Electrochemical Cell

## NRL Differential Calorimeter



$$P_{out} = a + b \cdot \Delta V$$

a= offset

b=gain

$\Delta V = \text{ddp peltier}$

## Observed Behaviors

Some differences were observed concerning Pd cathodes loaded above the threshold  $D/Pd = 0.9$ :

- 1) High power gain during the excess.
- 2) Low power gain during excess.
- 3) No excess.

The different behavior was related to some features of the samples ascribed to contaminants.

# Differences in Two Lots from the Same Producer

Spectrum of Contaminants in the Rough Materials (from the producer).

Both lots 99.95% purity

Ag 3	Al 5	As	Au 6	Be	B 5	Bi 10	Ca 2	Cd 2	Cl	Co 2	Cr 5	Cu 3	Fe 15
Ga	Ge	Hf	In	Ir 12	K	Li	Mg 2	Mn 2	Mo 3	Na	Nb	Ni 5	P
Pb 5	Pd Base	Pt 190	Rh 12	Ru 3	S	Sb 5	Si 30	Sn 10	Ta	Te	Ti	V	W
Zn 5	Zr 20	C	H	O	N	Y							

I lot: Reproducibility > 60%, Excess Power > 100%

Ag	Al 2	As	Au 3	Be	B	Bi	Ca	Cd	Cl	Co	Cr	Cu 3	Fe 3
Ga	Ge	Hf	In	Ir	K	Li	Mg	Mn	Mo	Na	Nb	Ni	P
Pb 4	Pd Base	Pt 80	Rh 31	Ru	S	Sb	Si 3	Sn	Ta	Te	Ti	V	W
Zn	Zr	C	H	O	N	Y							

II lot: Reproducibility < 20%, Excess Power < 20%



## Effect of Contaminants

Contaminants may act on:

- Grain size



Controls Stress and Mass Transfer

- Crystal orientation



Controls Kinetics and DL Capacitance

- Grain boundary

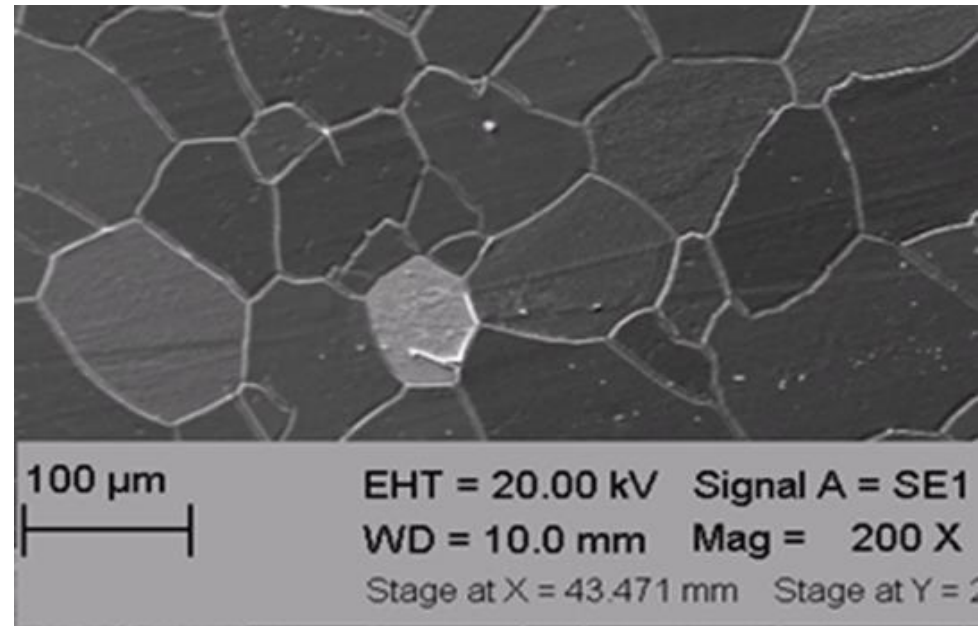
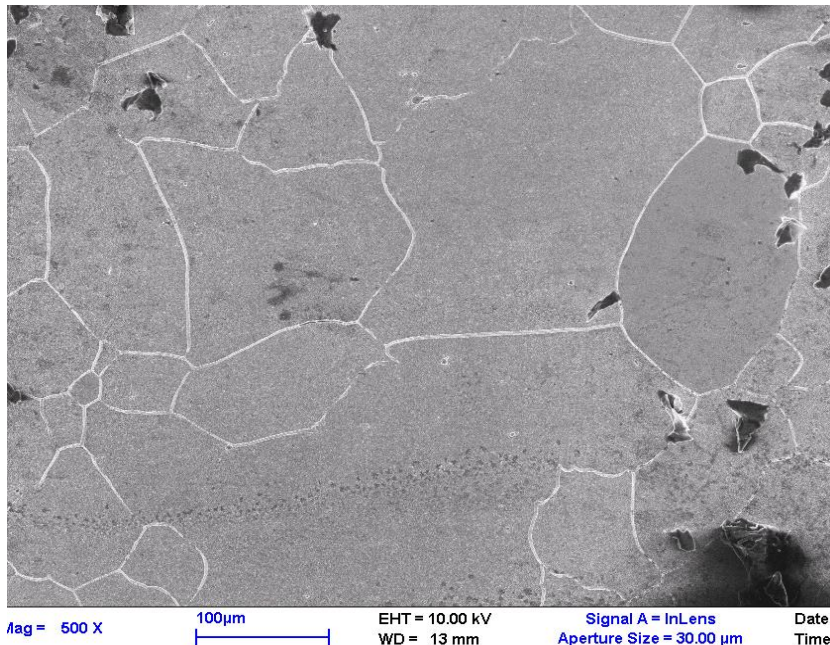


Controls Stress and Mass Transfer

- Surface treatment → Surface Morphology



Acts on Kinetics and DL Capacitance



Lot 1 From the same producer; same treatment Lot 2



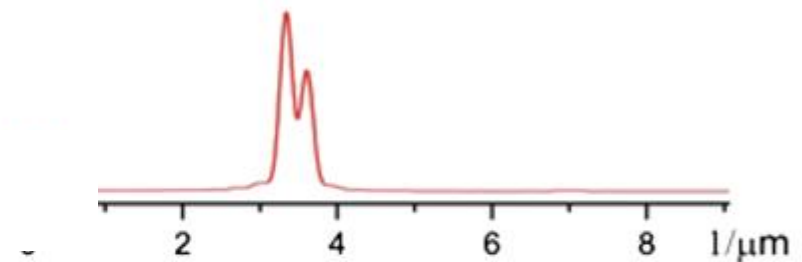
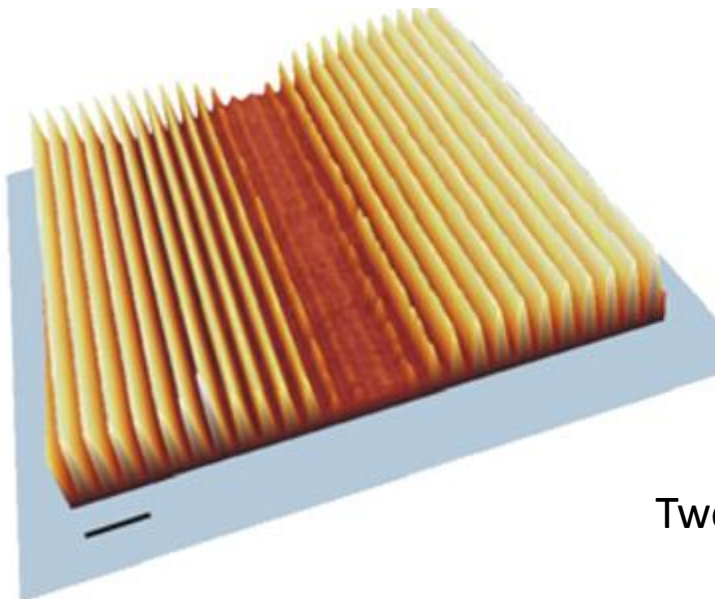
## Identified Conditions to Observe the Effect in PdD System

- 1) Appropriate metallurgy to achieve the loading threshold
- 2) Enhanced mass transfer
- 3)  $\langle 100 \rangle$  mostly oriented material
- 4) Appropriate surface morphology

# Surface

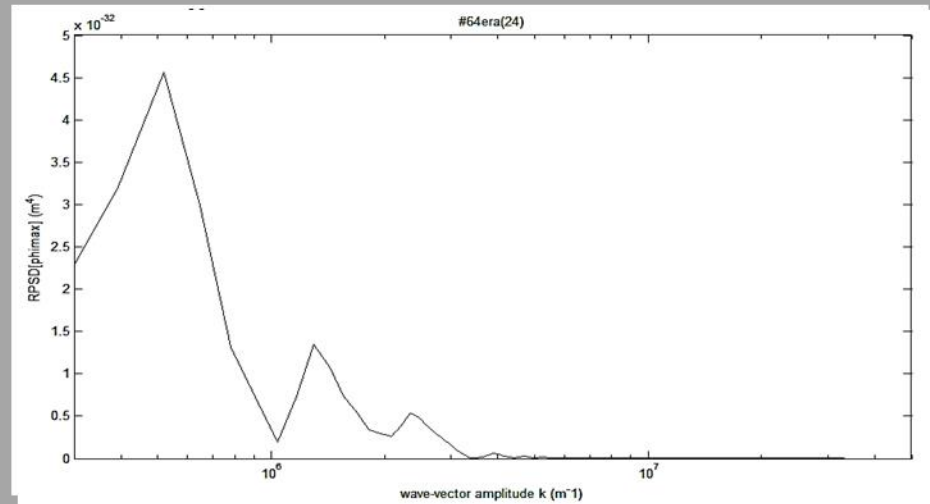
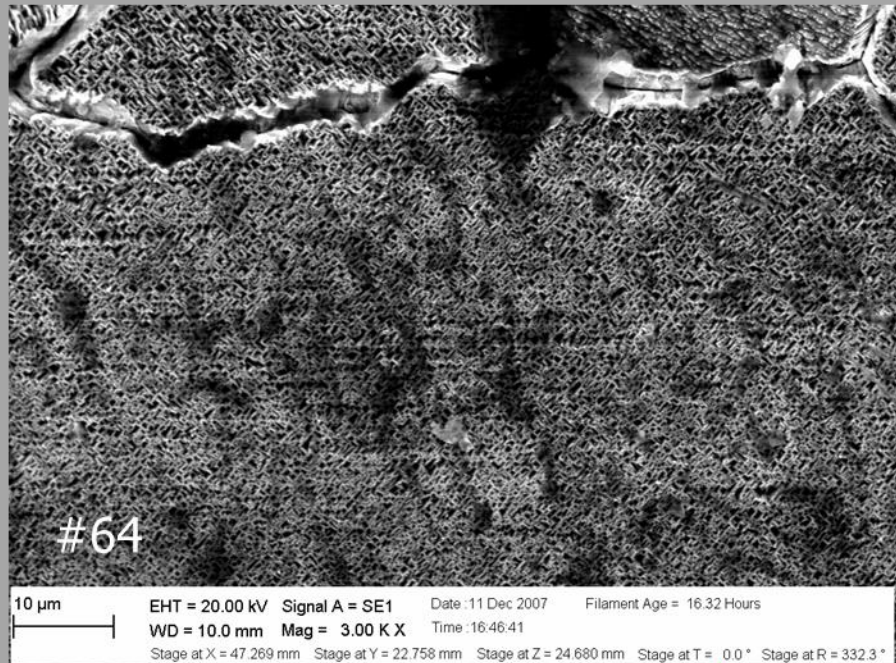
- Crystal orientation and specific contaminants modify the effect of the chemical etching leading a different surface status.
- The surface morphology is acting on the interface electrochemical structure.

Power Spectral Density Function (PSDF) has been selected as surface merit figure.

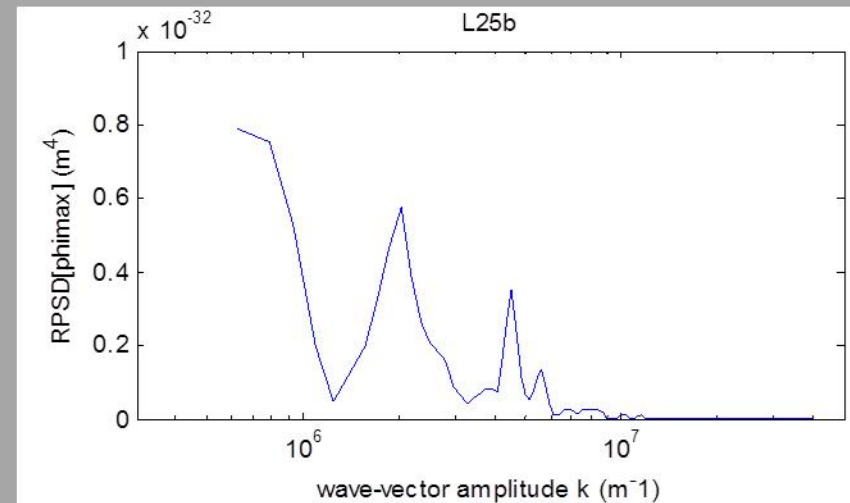
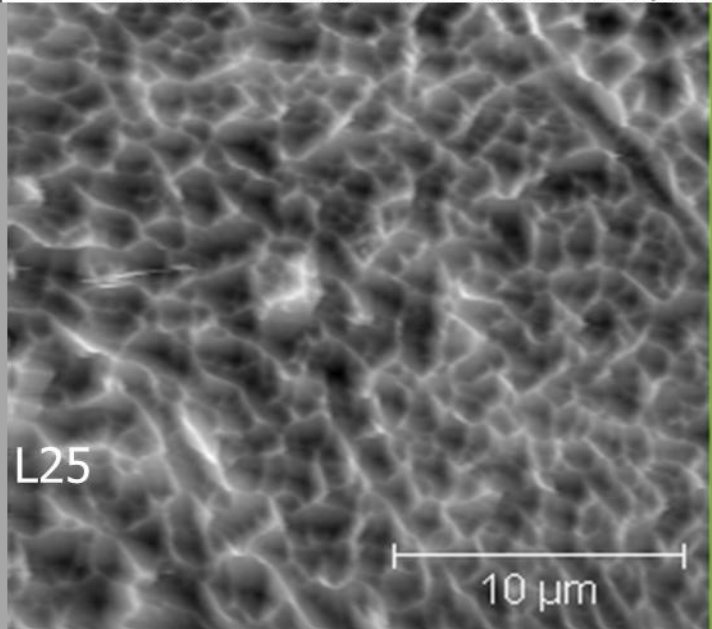


Two periodicity system

# Typical Surface Morphology (after Etching) giving Excess of Heat



PSDF of sample #64 **producing 3500% excess of heat.**



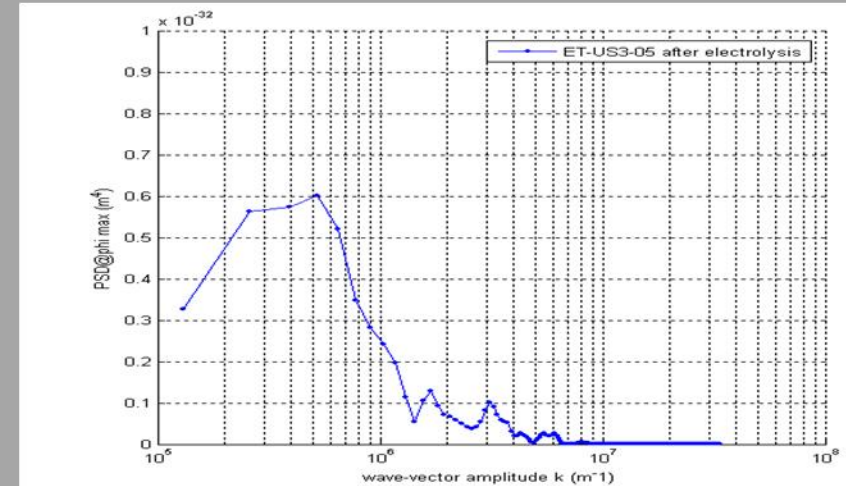
PSDF of sample L25 **producing up to 250% excess of heat.**



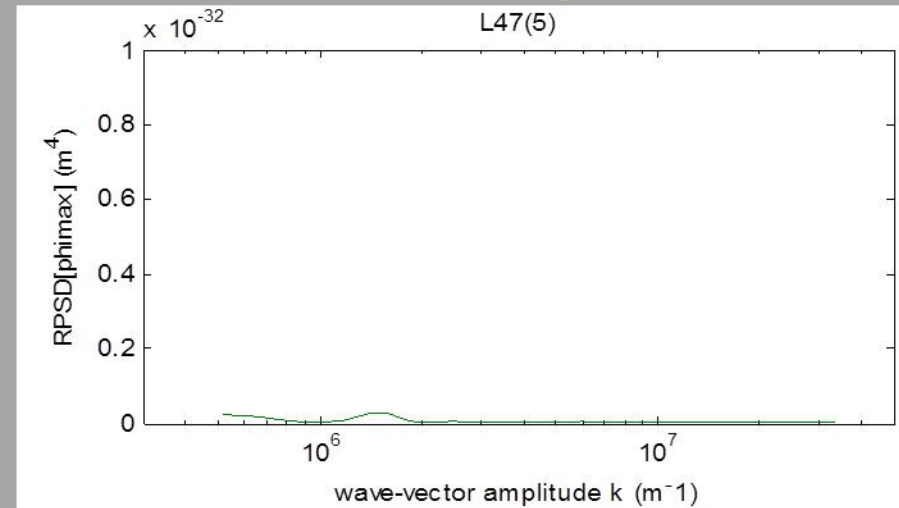
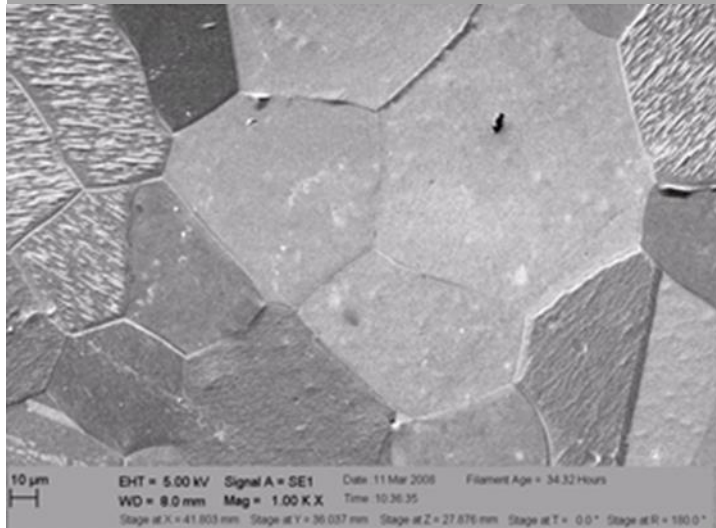
The larger the PSDF amplitude the larger the excess of power



10  $\mu$ m EHT = 20.00 kV Signal A = SE1 Date : 11 Dec 2007 Filament Age = 15.48 Hours  
WD = 10.0 mm Mag = 3.00 K X Time : 15:56:45  
Stage at X = 44.870 mm Stage at Y = 23.649 mm Stage at Z = 24.680 mm Stage at T = 0.0° Stage at R = 52.9°



PSDF of sample ET-UTS-05 **producing 25% excess of heat under electrolysis.**



PSDF of a sample **not producing excess of heat**

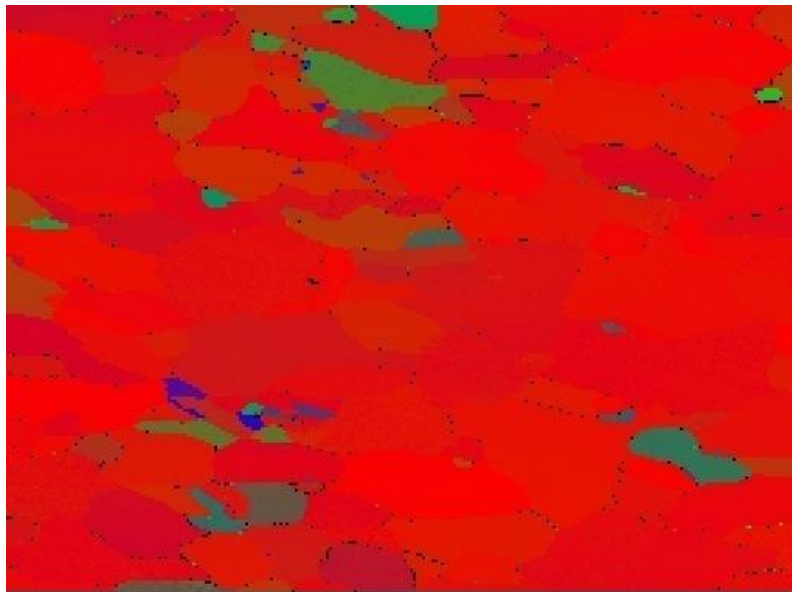
Surface morphology results to be a fourth condition

Now we play with contaminants

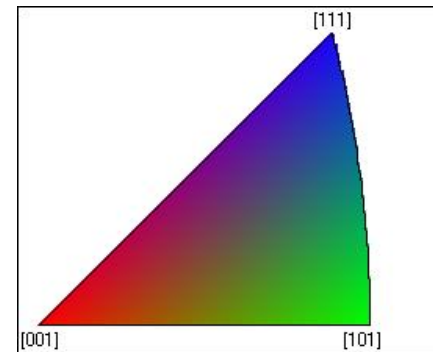
## Producing an Active Material

The experimental evidences allowed to produce a material with characteristics close to the ones above described.

A lot of Pd having a spectrum of contaminants approaching the lot 1 was undergone to the treatment leading to: dominant  $\langle 100 \rangle$  orientation and an appropriate metallurgy.

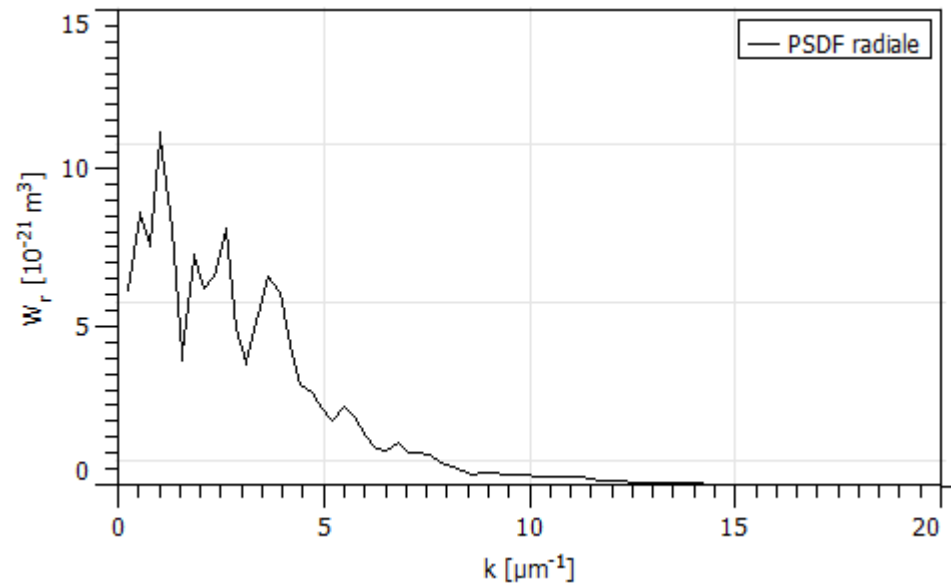
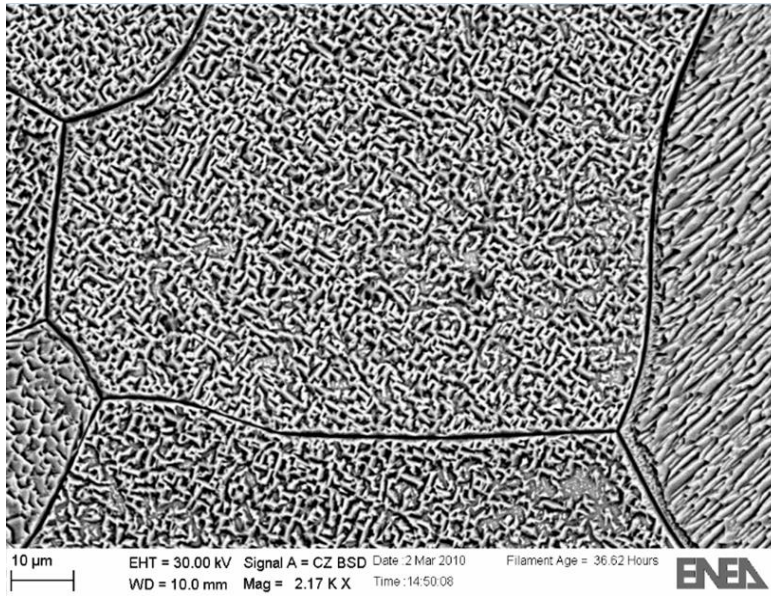


Sample Normal

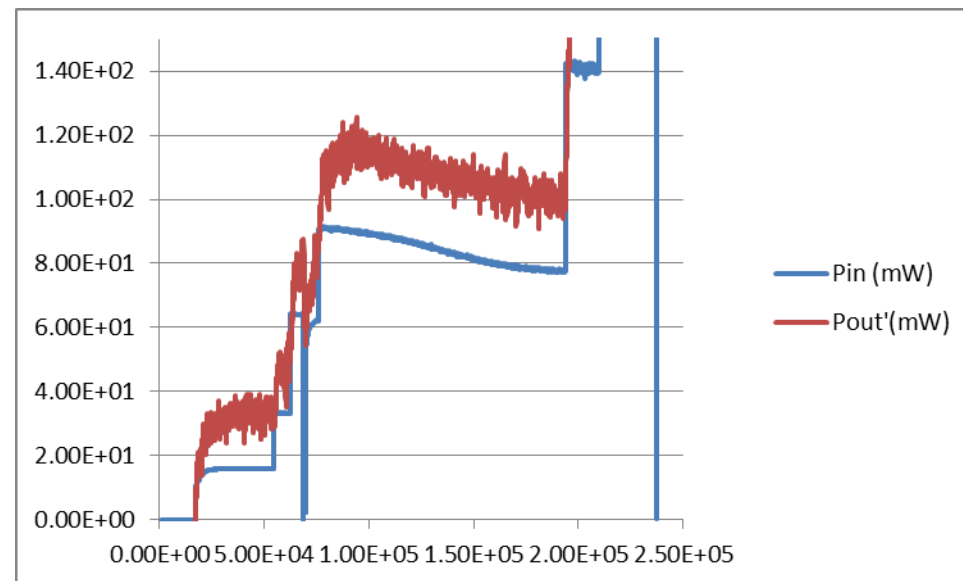
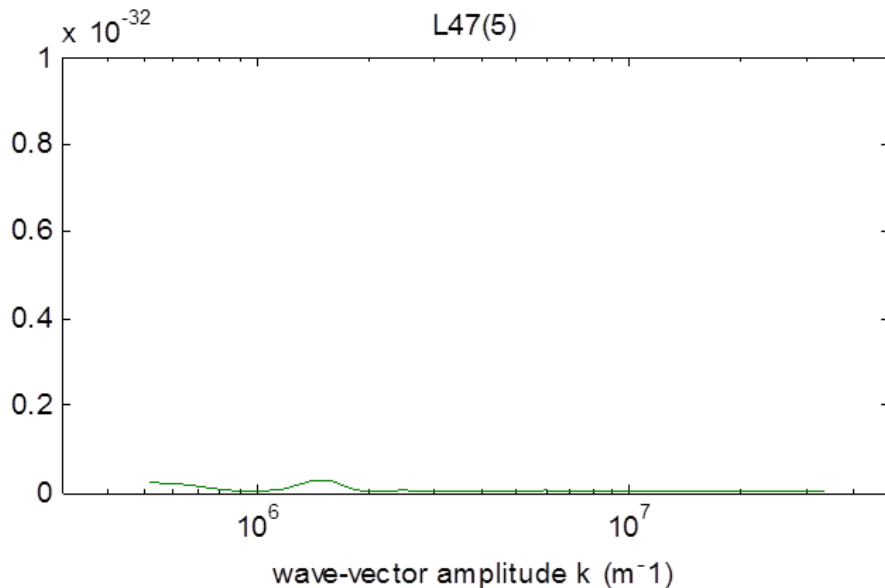


# Palladium was doped by Platinum: effect on the surface

L66 (120-160) Platinum added



up to 25-30% excess

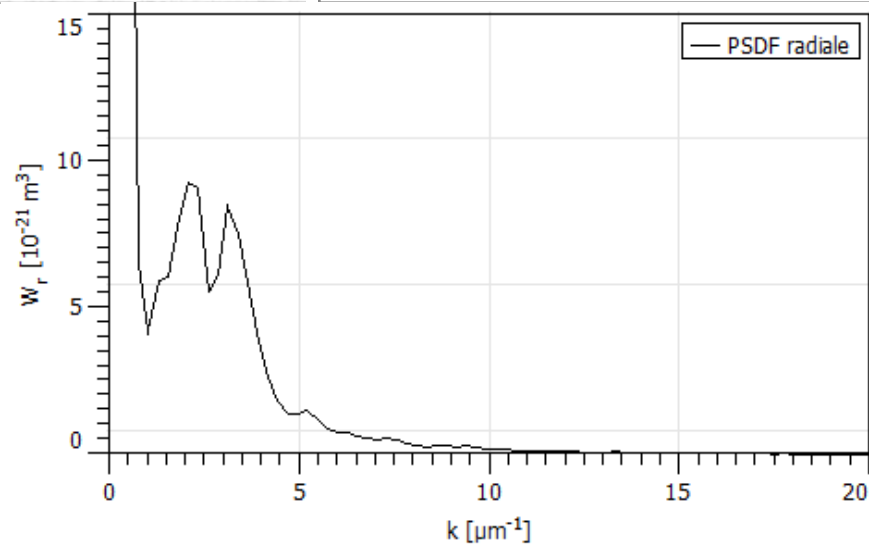
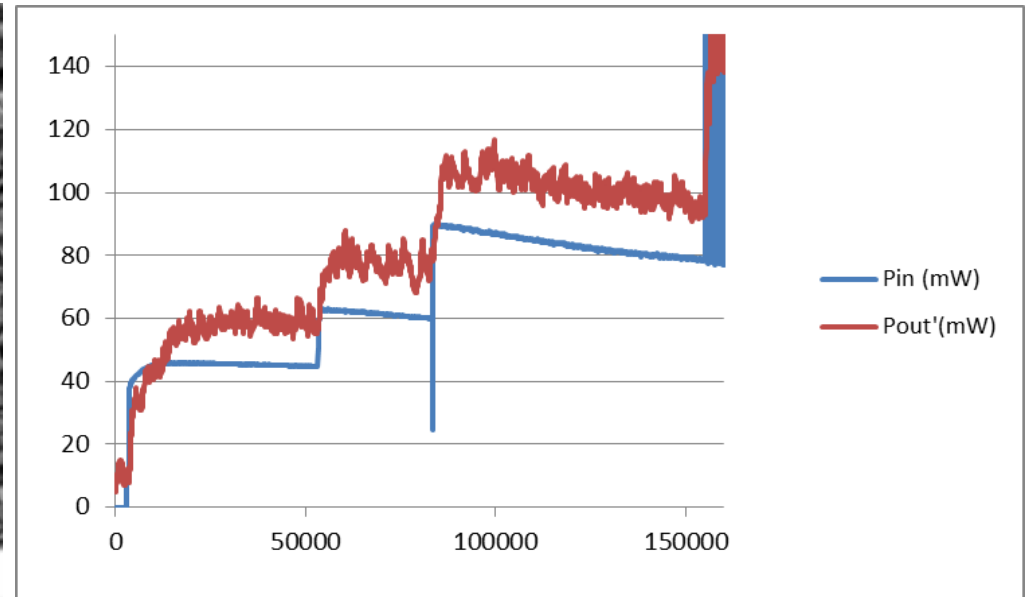
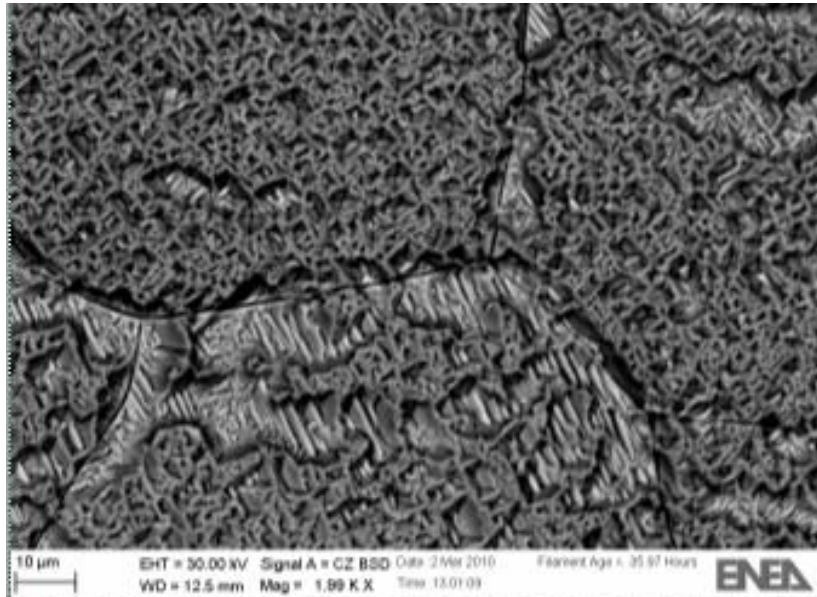




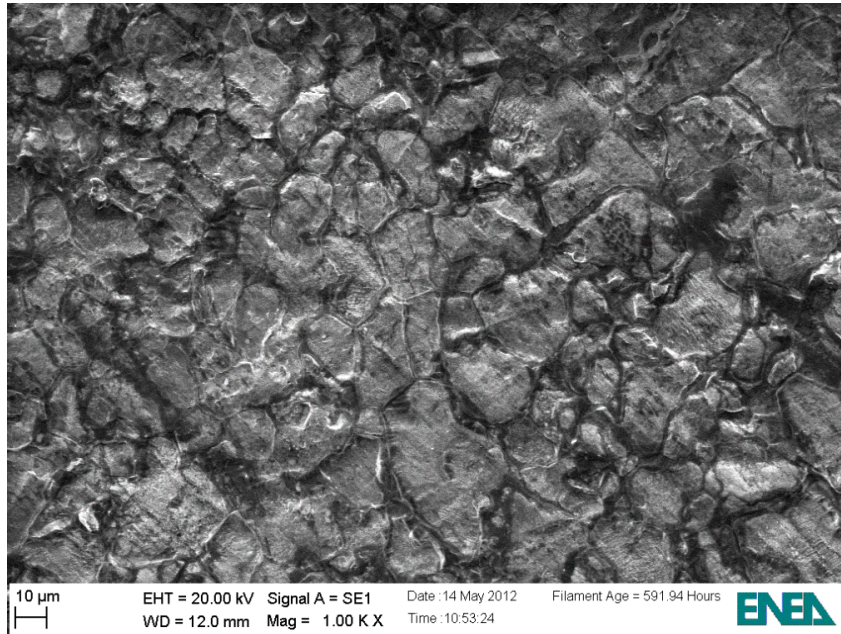
Again

L66 (160-200) Platinum added

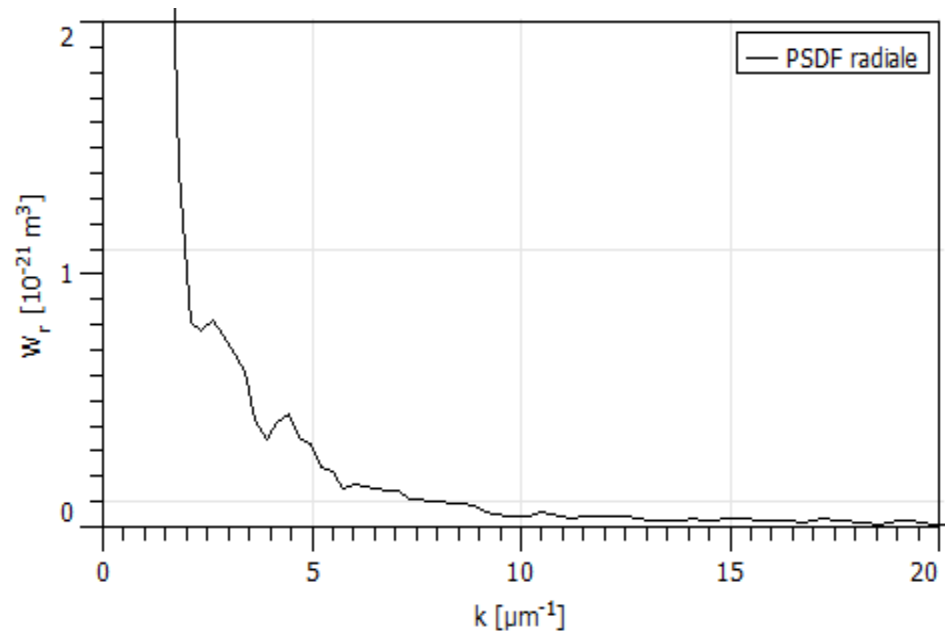
up to 25-30% excess



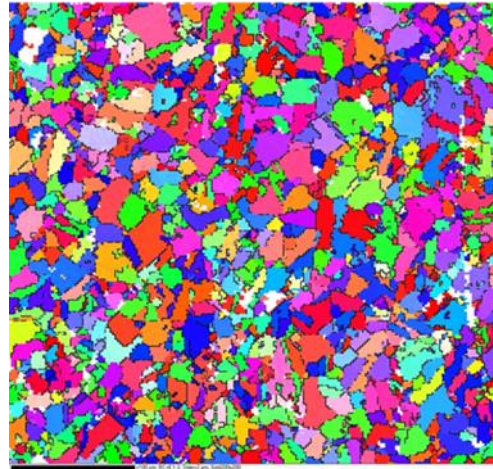
## Palladium Rhodium Alloy produced at ENEA by using NRL protocol



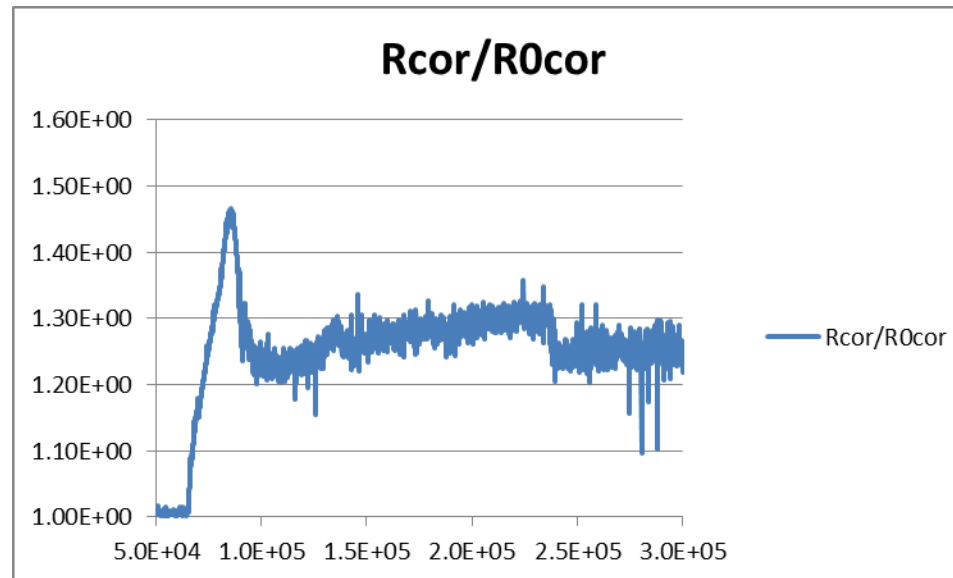
L119 (20-60) Before



L119 (20-60) PSD from AFM

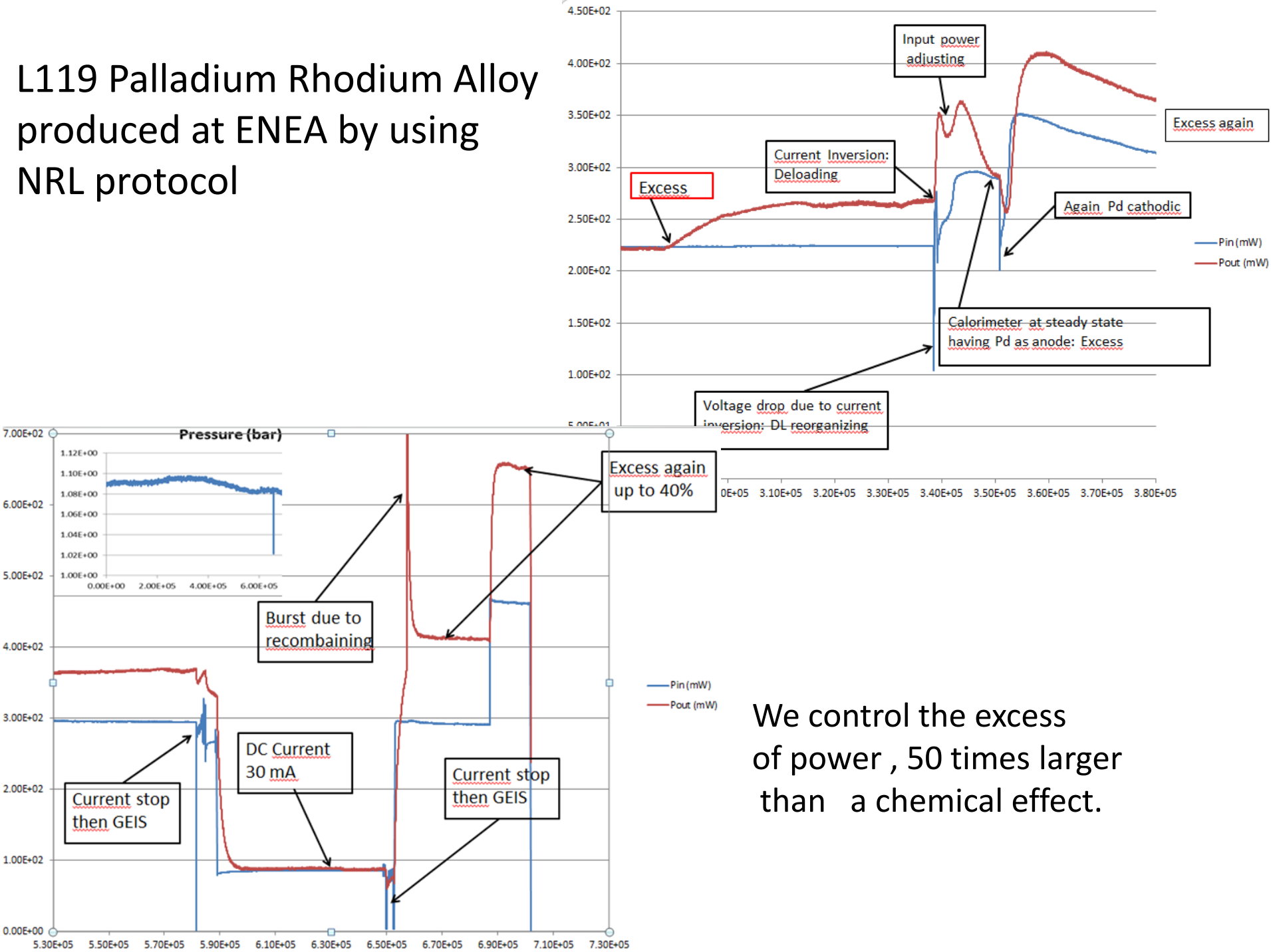


EBSD reveals a scattered orientation of crystals and small size grains



Loading is high and fast

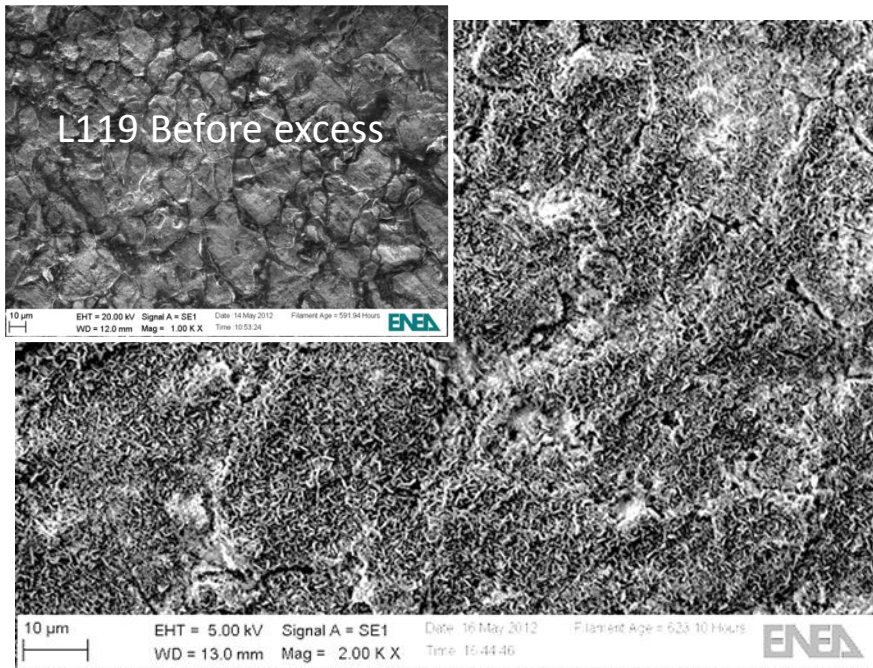
# L119 Palladium Rhodium Alloy produced at ENEA by using NRL protocol



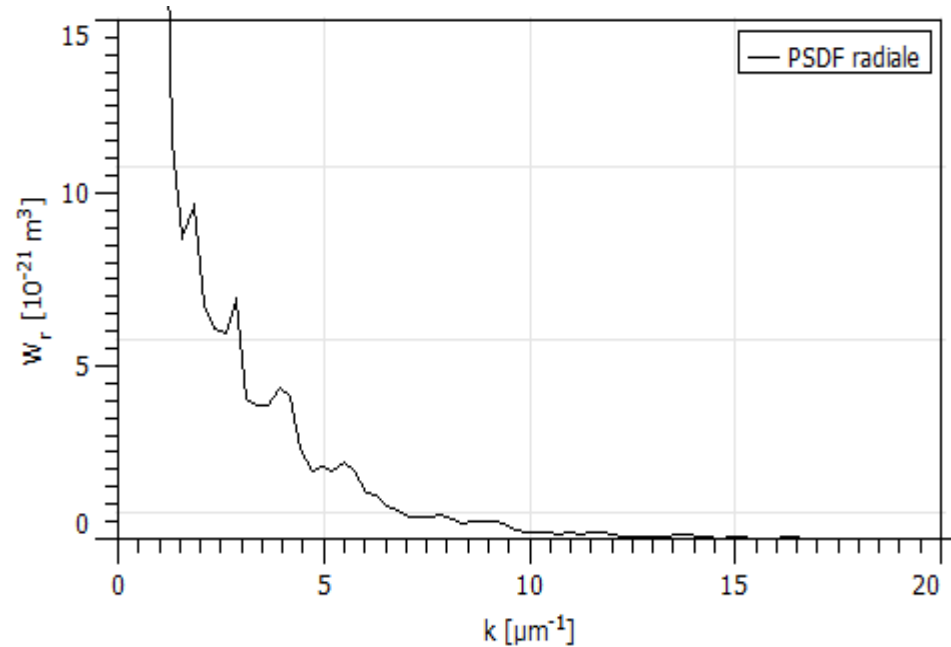
We control the excess of power , 50 times larger than a chemical effect.



## L119 as was During the Excess



Surface

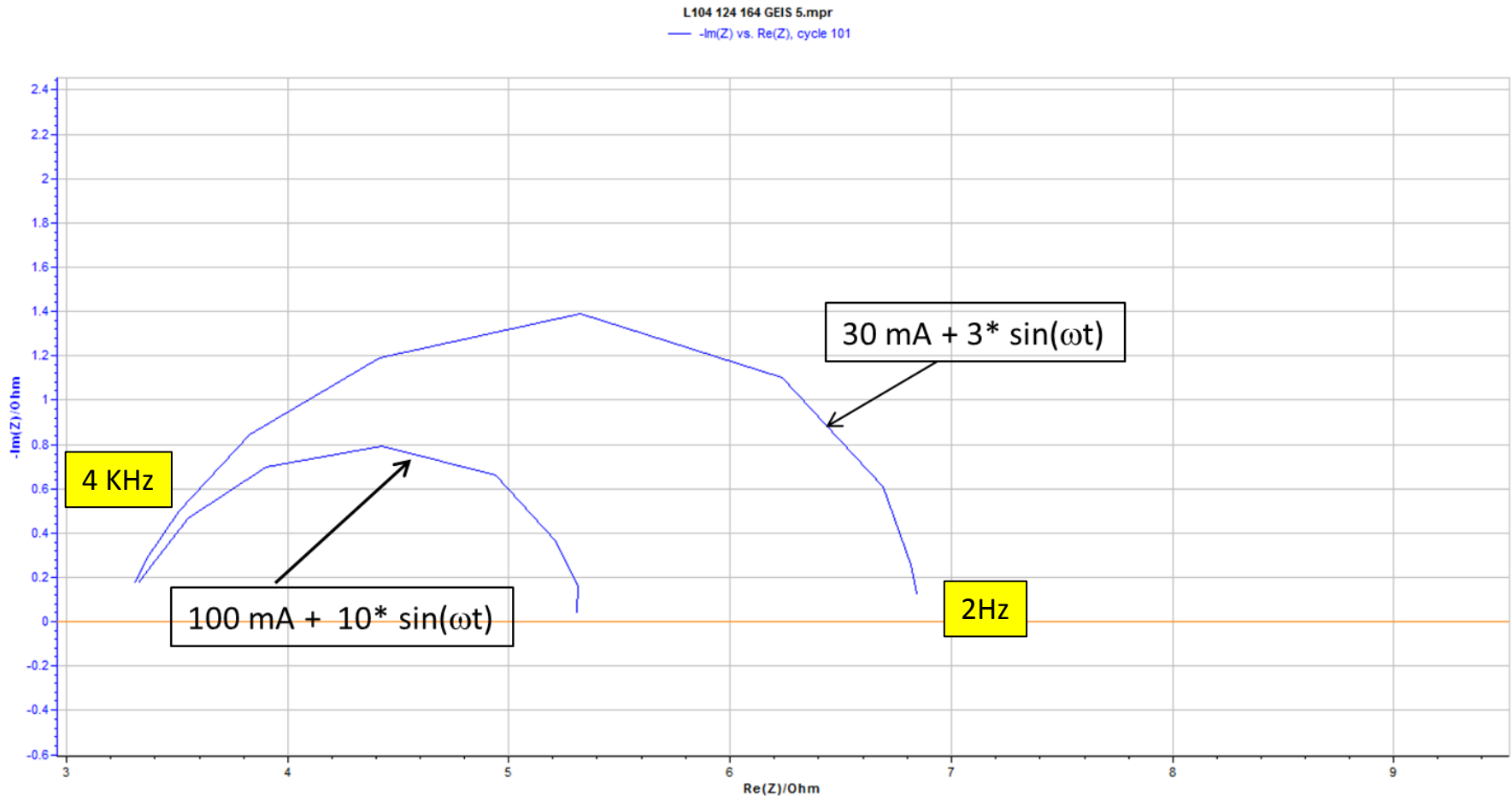


PSDF

Cu, Fe, Si and Pt have been identified as contaminants on the electrode surface after the electrolysis.

Let see if we may extract any additional information from  
GEIS (Galvanostatic Electrochemical Impedence Spectroscopy)  
performed during the experiment.

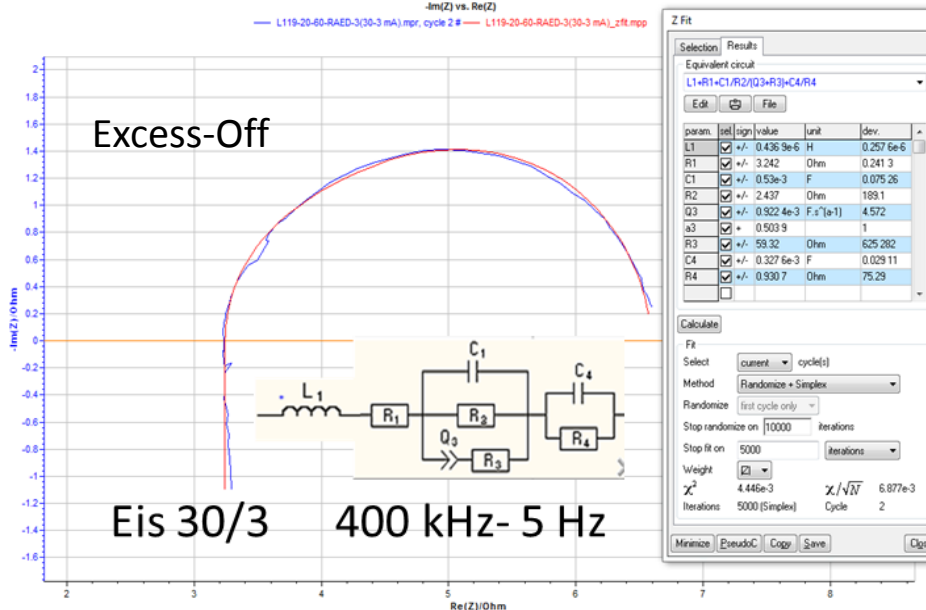
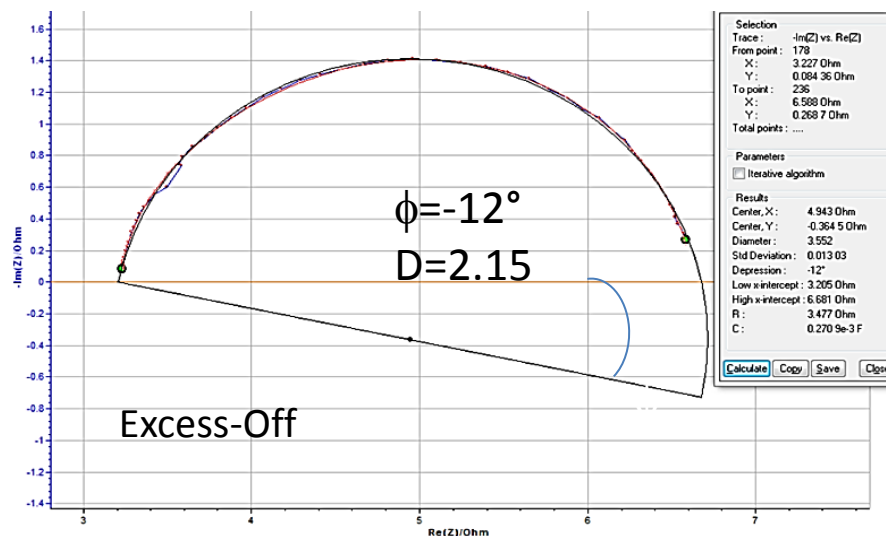
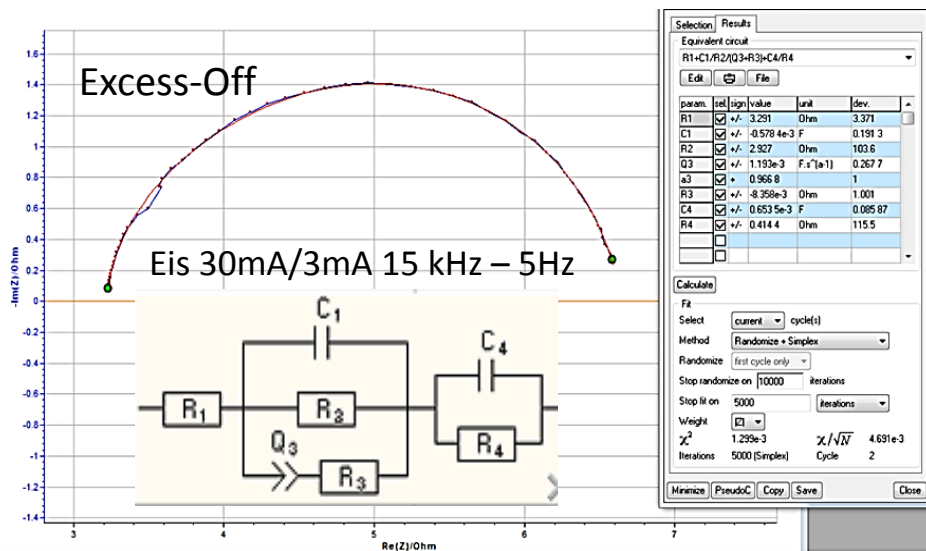
# Typical GEIS Performed with Inactive Pd Foil



GEIS at two DC current levels



# In Situ Electrochemical Impedance Spectroscopy on Sample L119(20-60) Excess-off

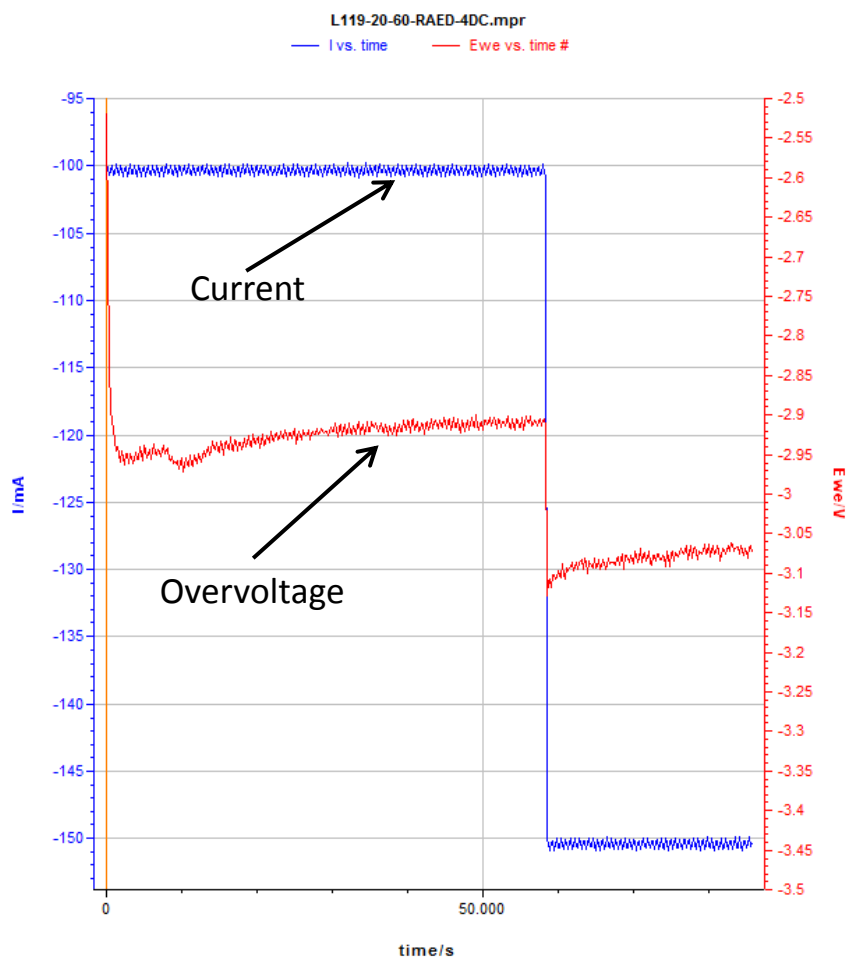


Depression angle

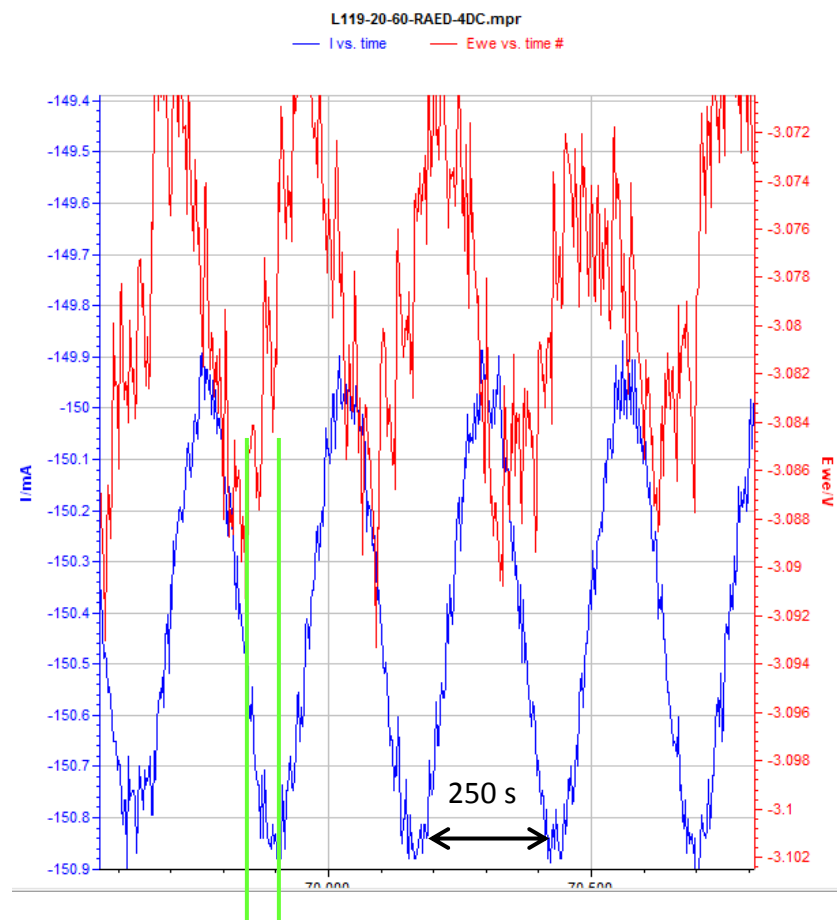
$$\phi = (1-n) \cdot 90 \quad D = 1 + 1/n$$

D Fractal dimension

## Current and overvoltage behavior during 40% excess

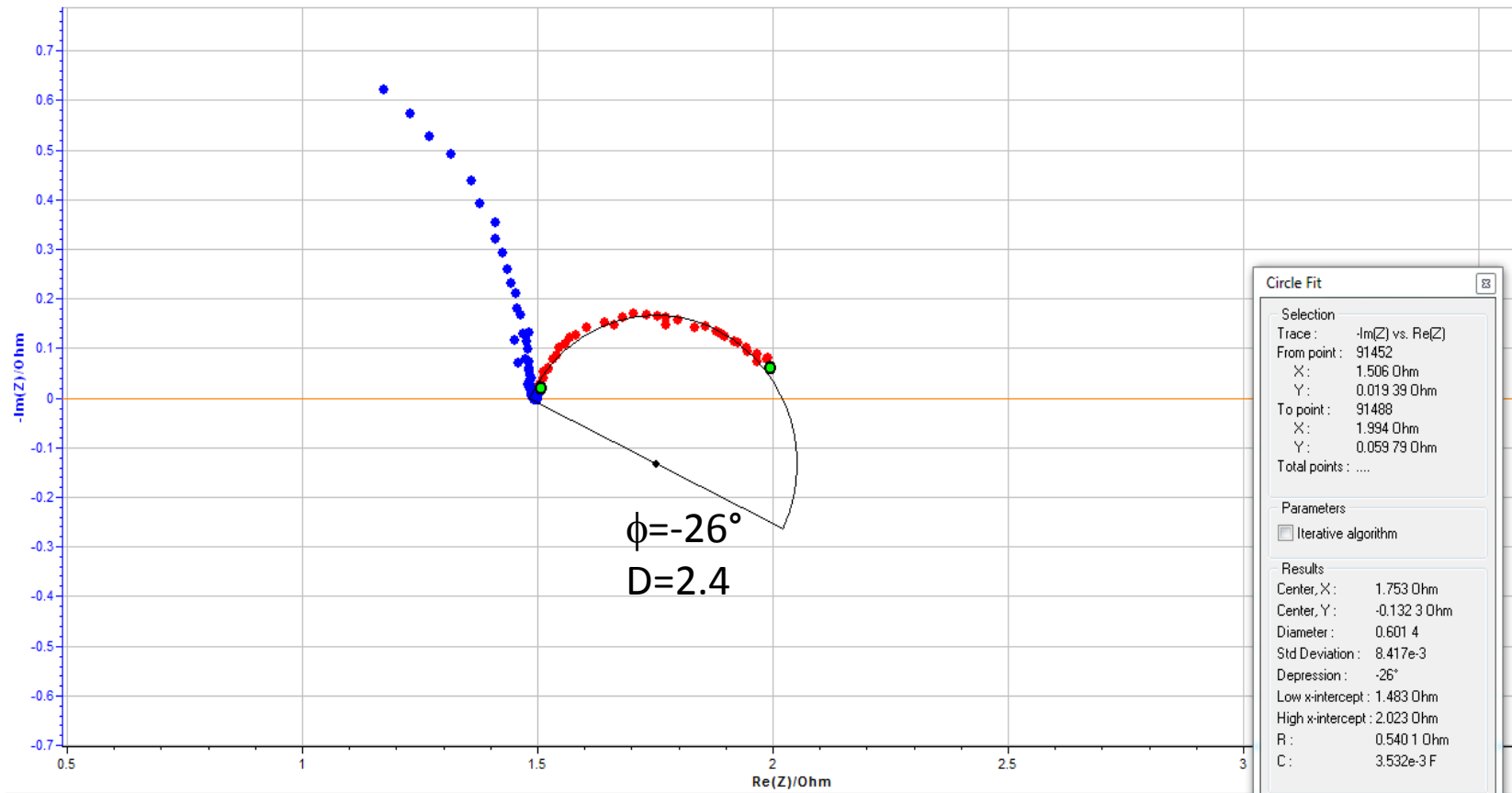


Zooming

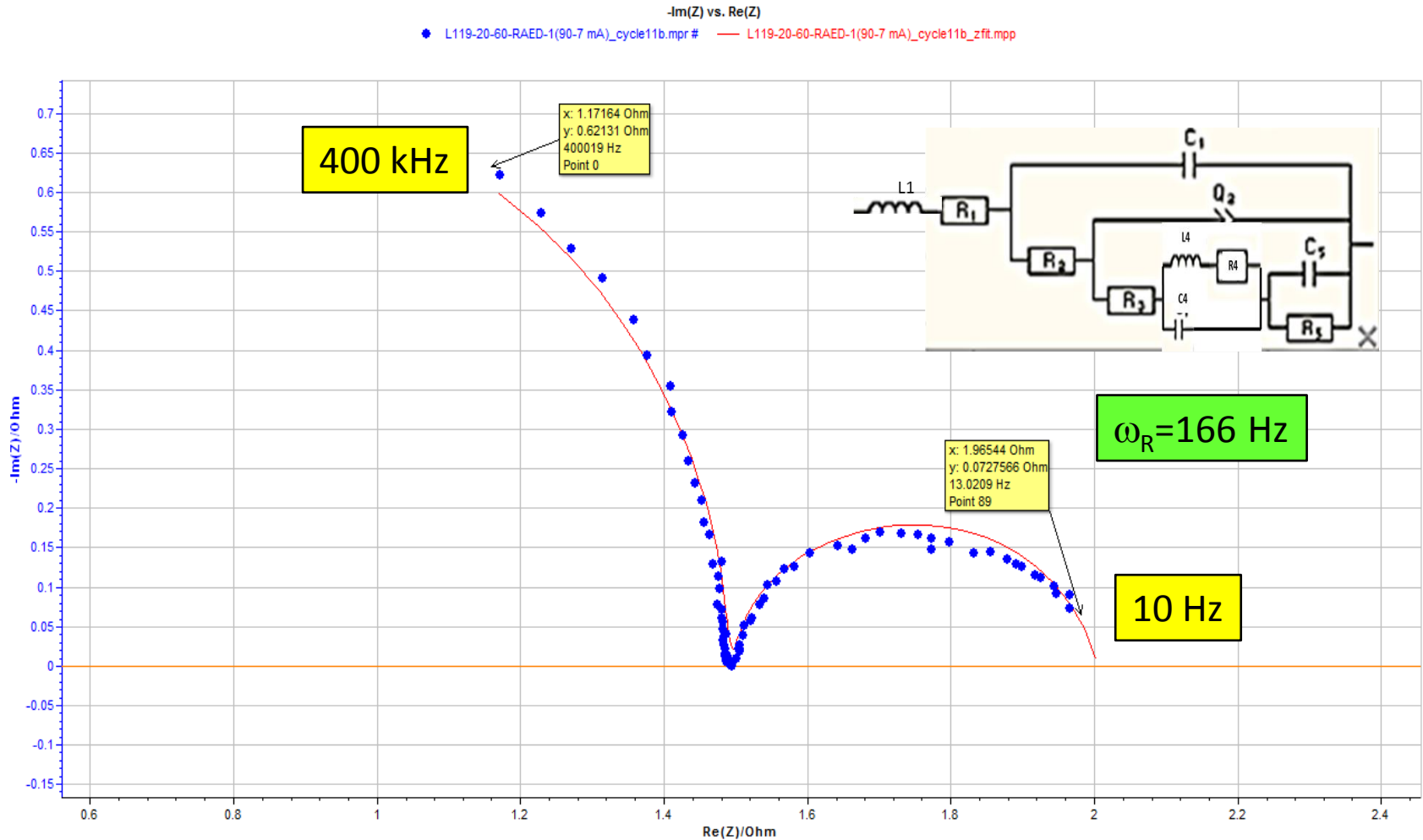


Note: Current ripple has been always observed in galvanostatic mode during significant excesses

Something was changing at the interface during the excess !!

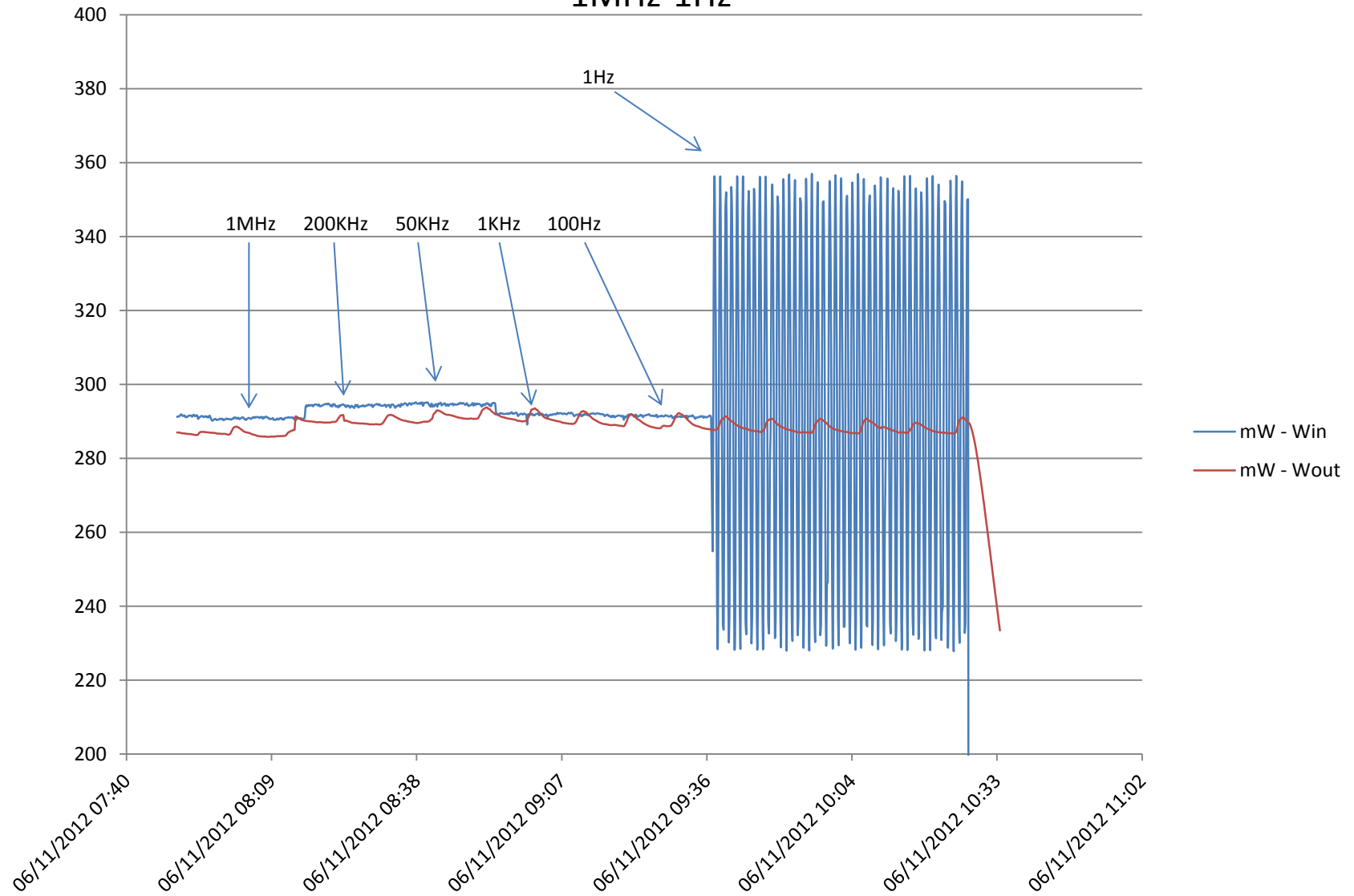


# In Situ Electrochemical Impedance Spectroscopy on Sample L119(20-60) Excess-on



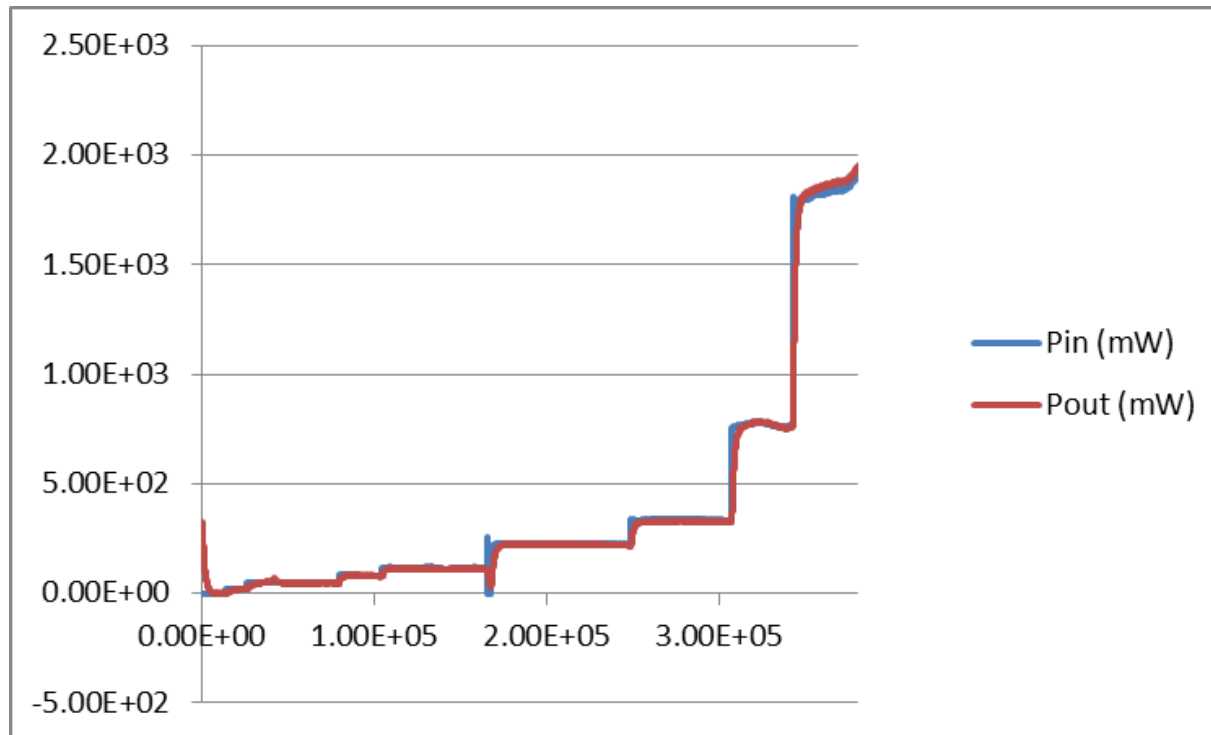
A wide range of resonating frequencies is possible

**Calorimeter check:** Pin-Pout by applying  $100\text{mA} \pm 20\text{mA}$  in the frequency range  
**1MHz-1Hz**



## Sample L119-140-180

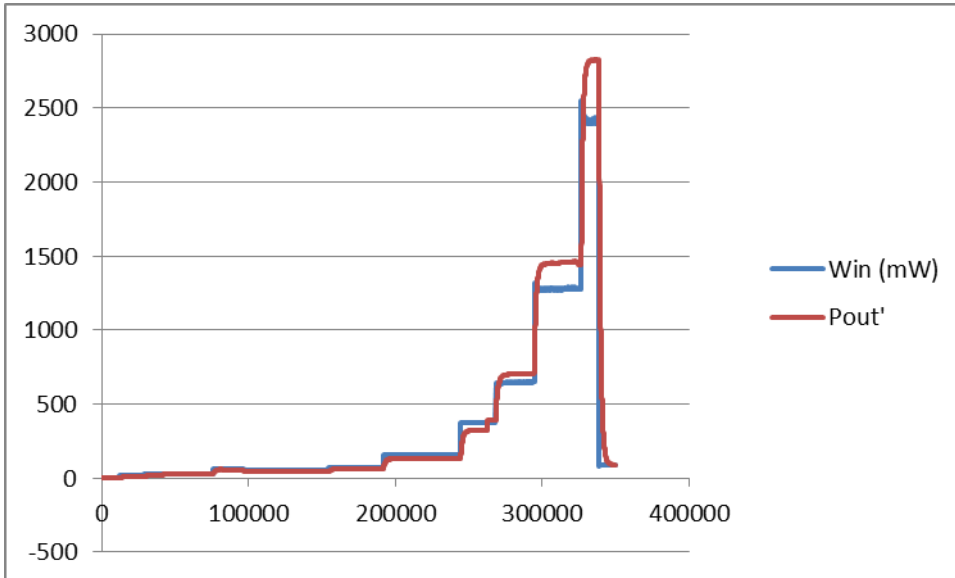
Sample L119-140-180 was inactive, no evidence of specific contaminants on the surface



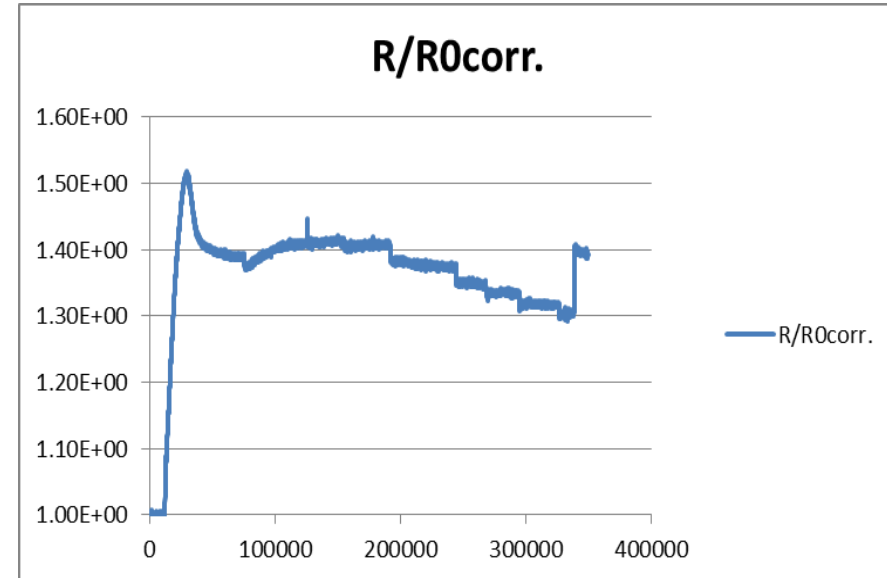
Pin-Pout Sample L119-140-180

Again

L124(50-90) Excess



Pin - Pout L124(50-90)



Normalized resistance L124(50-90)

Loading is high and fast

Identified contaminants : same as L119(20-60)

This sample was the 'most close' to L119(20-60)



## Conclusions

Material features , related with the occurrence of the effect, have been identified:

- 1) loading threshold, 2) loading dynamics, 3) grain-grain boundary size,
- 4) surface morphology, 5) crystal orientation (Pd).

A specific role of some contaminants has been also identified.

In situ GEIS revealed a dramatic changing of the electric structure of the interface : resonating circuits components turn out during excess.

**Material status** is the key to observe the effect. **Material science** is the key to understand it, since some material characteristics support some processes rather than others.

By applying the scientific method future work should be oriented towards the definition of the effect rather than its demonstration.

*Thank You*

This work has been supported by National Instruments and a specific work is in progress to develop new instruments to improve the study.