

Anomalous Heat Generation in Charging of Pd Powders with High Density Hydrogen Isotopes (I) Results of absorption experiments using Pd powders

Y. Sasaki¹, A. Kitamura¹, Y. Miyoshi¹, T. Nohmi¹, A. Taniike¹, A. Takahashi²,
R. Seto², and Y. Fujita⁴

¹ Division of Marine Engineering, Graduate School of Maritime Sciences, Kobe University Higashinada-ku, Kobe 6580022, Japan

² Technova Inc, Chiyoda-ku, Tokyo 1000011, Japan

⁴ To confirm heat and He generation by deuterium (D) absorption in nano-sized Pd powders reported by Arata and Zhang [1], and to investigate the underlying physics, we have installed a twin system of double structured vessels to perform flow calorimetry during D₂ or H₂ absorption by a variety of micronized Pd samples.

The first-stage experiments are described in detail in ref. [2]. The evolution of pressure and temperature after introduction of 1-MPa D₂/H₂ gas was divided into two phases. The first phase is zero-pressure interval, and in the second phase, pressure increases up to the stationary value. When D₂ gas was used with Pd-black, apparent excess heat production in the second phase was implied, although temperature oscillations and drift were too large to confirm the result.

Then in the second stage, the system was modified to improve the accuracy: The heat capacity of the reaction vessel was decreased, while increasing the mass of the test sample, to minimize the time constant of the calorimeter and maximize the sensitivity.

Nano-sized powders of mixed Pd and Zr oxides fabricated by Santoku Corporation, Kobe, Japan, have been used to reveal their interesting and exciting characteristics. In the 1st phase, D-gas charge gave 20~90 % excess heat compared to H-gas charge. In the 2nd phase, significant excess heat (about 2 kJ/g-Pd) for D-gas charge was observed, in contrast to near zero level output for H-gas charge. We will further examine the dependence of the anomalous excess heat on the experimental conditions such as the gas flow rate and the sample temperature. The anomalies and the possible mechanisms will be discussed in more detail in the succeeding presentation [3].

- [1] Y. Arata and Y. Zhang: The special report on research project for creation of new energy, J. High Temperature Society, 2008, No. 1; Y. Arata, and Y. Zhang: *Condensed Matter Nuclear Science, Proc. 12th Int. Conf. on Cold Fusion* (ed. A. Takahashi, Y. Iwamura, and K. Ota, World Scientific, 2006) pp.44-54.
- [2] T. Nohmi, Y. Sasaki, T. Yamaguchi, Taniike, A. Kitamura, A. Takahashi, R. Seto, and Y. Fujita: Basic research on condensed matter nuclear reaction using Pd powders charged with high density deuterium, Proc. ICCF14 (Washington DC, Aug. 10-15, 2008).
- [3] A. Takahashi, A. Kitamura, Y. Sasaki, Y. Miyoshi, T. Nohmi, A. Taniike, R. Seto, and Y. Fujita: Anomalous Heat Generation in Charging of Pd Powders with High Density Hydrogen Isotopes, (II) Discussions on Experimental Results and Underlying Physics, this meeting.

Anomalous Heat Generation in Charging of Pd Powders with High Density Hydrogen Isotopes

(I) Results of absorption experiments using Pd powders

Y. Sasaki A. Kitamura, T. Nohmi, Y. Miyoshi, and A. Taniike
*(Division of Marine Engineering, Graduate School of Maritime
Sciences, Kobe University)*

A. Takahashi, R. Seto, and Y. Fujita
(Technova Inc.)

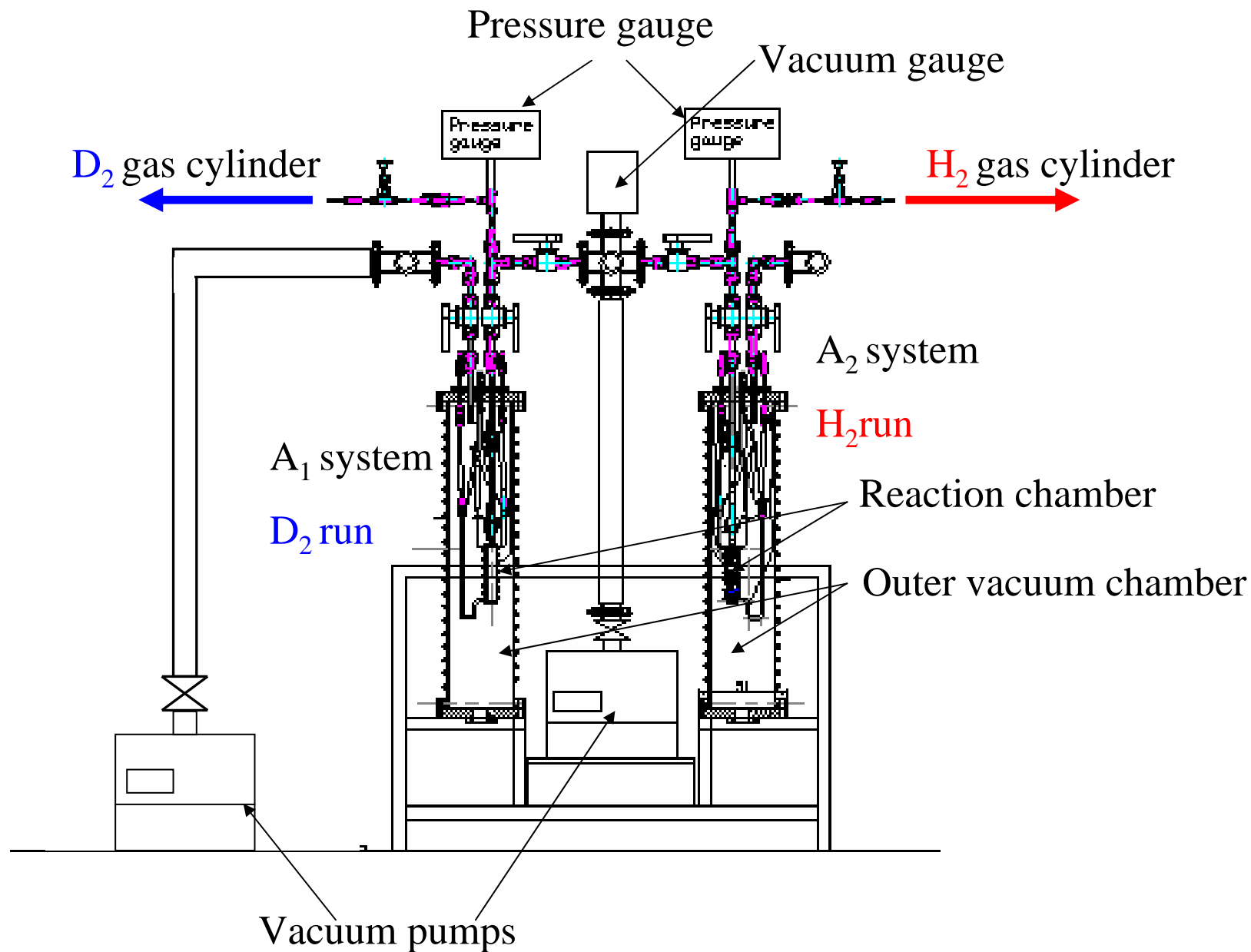
Aim

It has been reported in ref. [1] that charging of highly pure D2 gas into Pd nano-powders in the form of Pd/ZrO2 nano-composite contained in a stainless-steel vacuum vessel has induced significant excess heat.

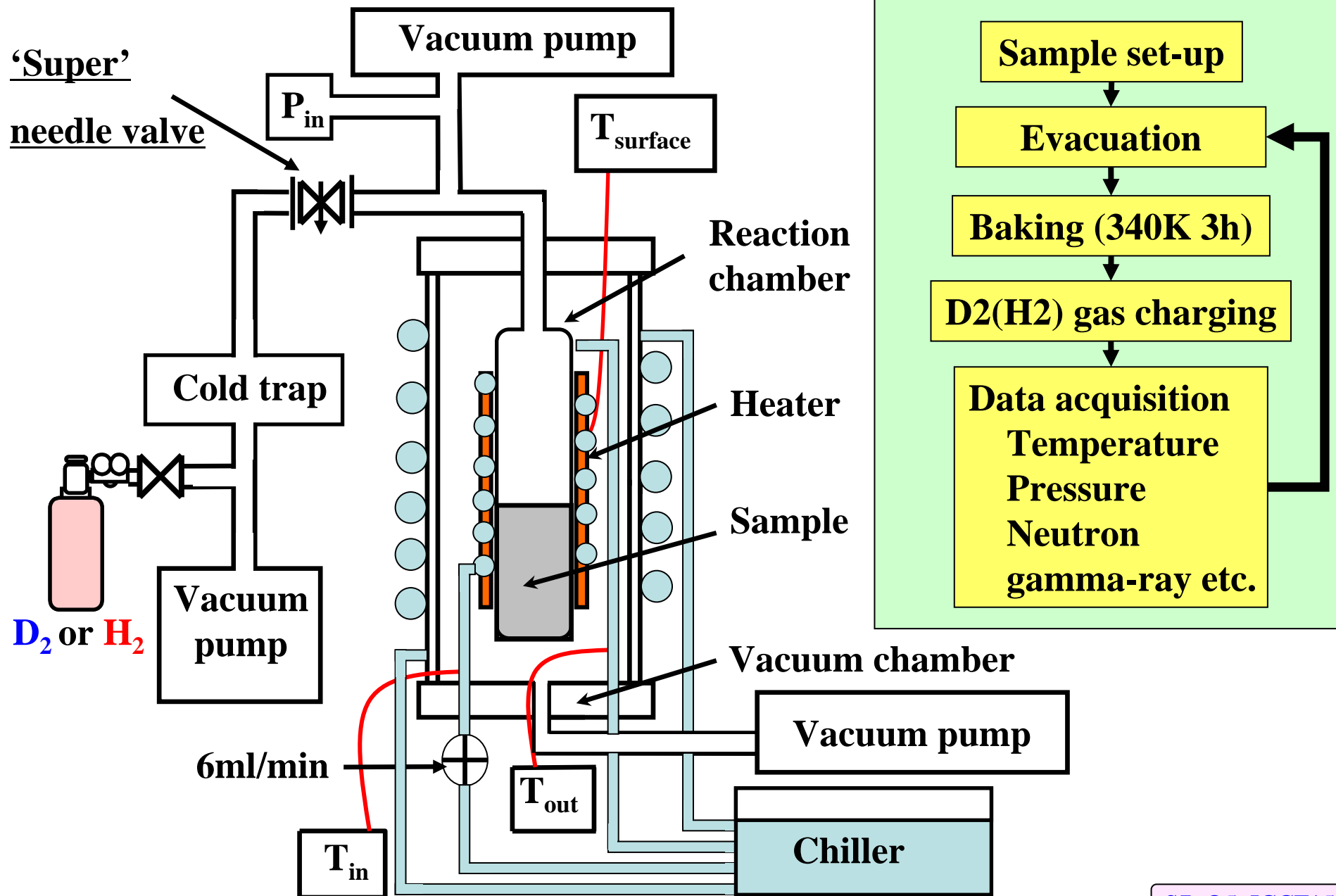
we have constructed an experimental system to confirm the phenomenon of heat and ^4He generation by calorimetry and investigate the underlying physics.

[1] Y. Arata, et al.; The special report on research project for creation of new energy, J. High Temperature Society, No. 1. 2008.

Reduced view of the twin system A1A2



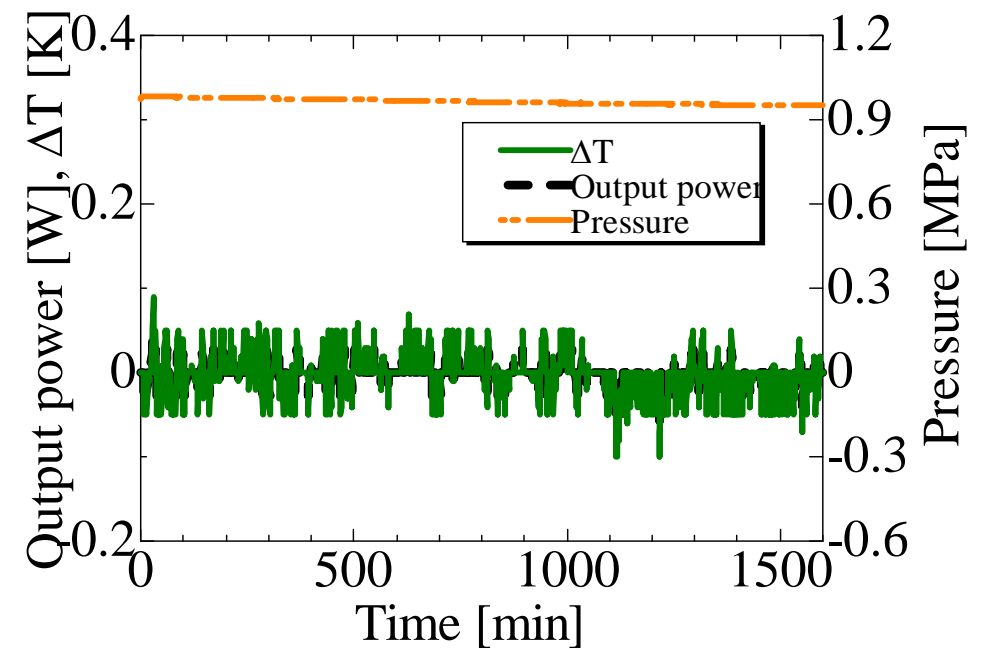
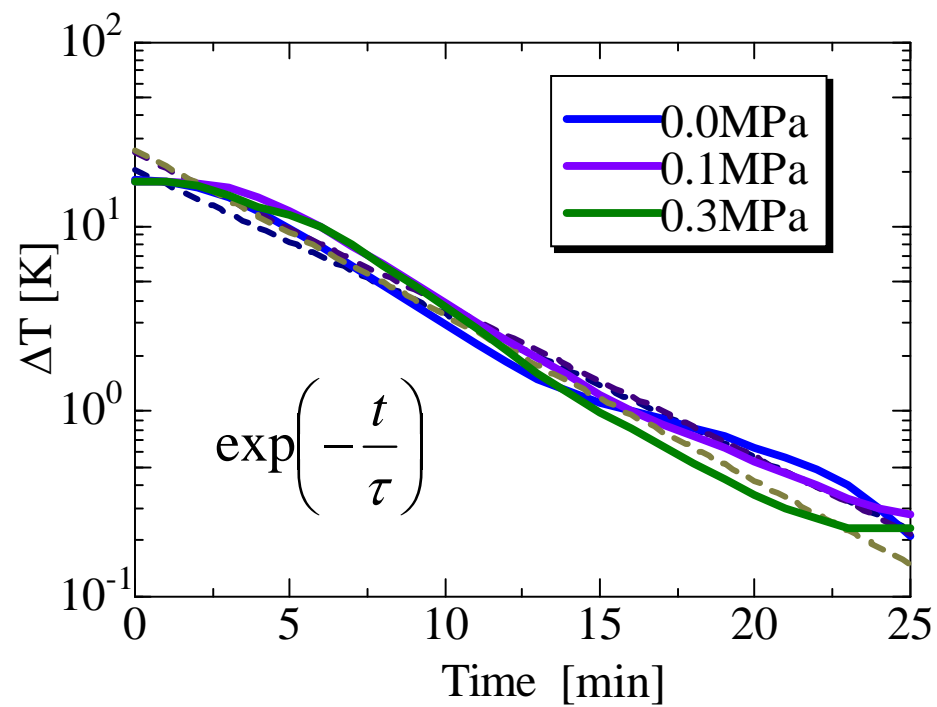
Functional view of the A₁ A₂ system



Performance of calorimetry

Time resolution : 5 min

Accuracy : ± 14 mW



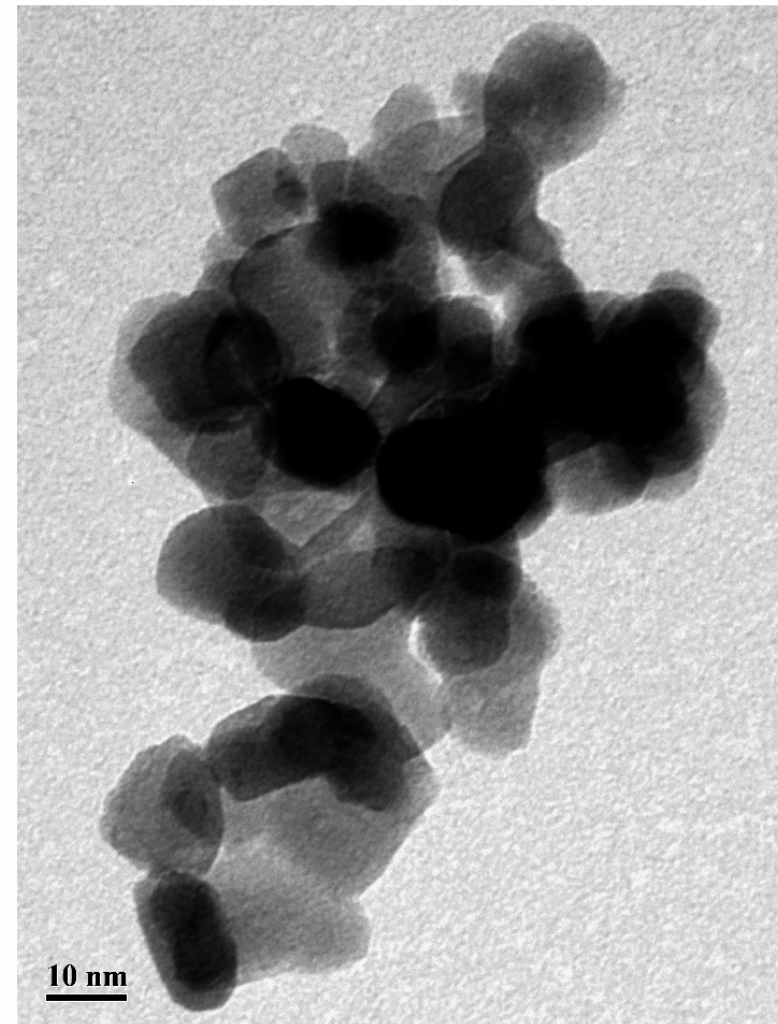
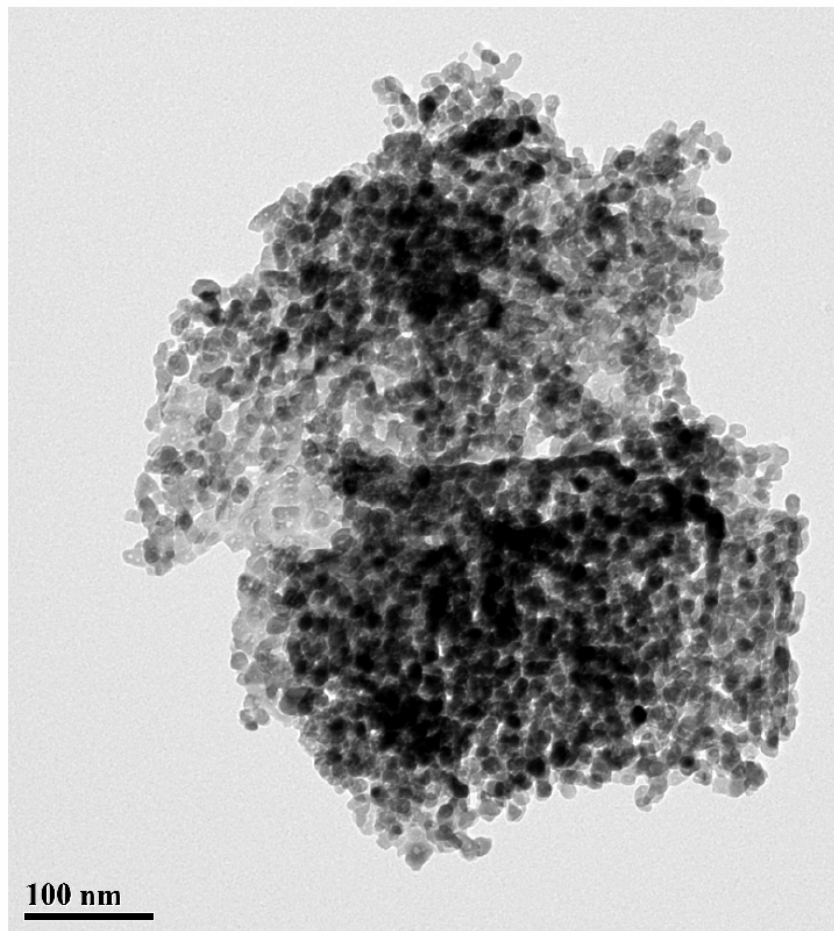
Samples

- ϕ 100 nm Pd (**PP**); This is a Pd powder diameter of particle is 100 nm, purity is 99.5%.
- Pd-black (**PB**); This is a 300 mesh powder and purity is 99.9%
- Santoku Pd (**PZ**); This is a nano-sized(**8 nm** and **10.5 nm**) powder of mixed-oxides of Pd and Zr (fabricated by Santoku Corporation)

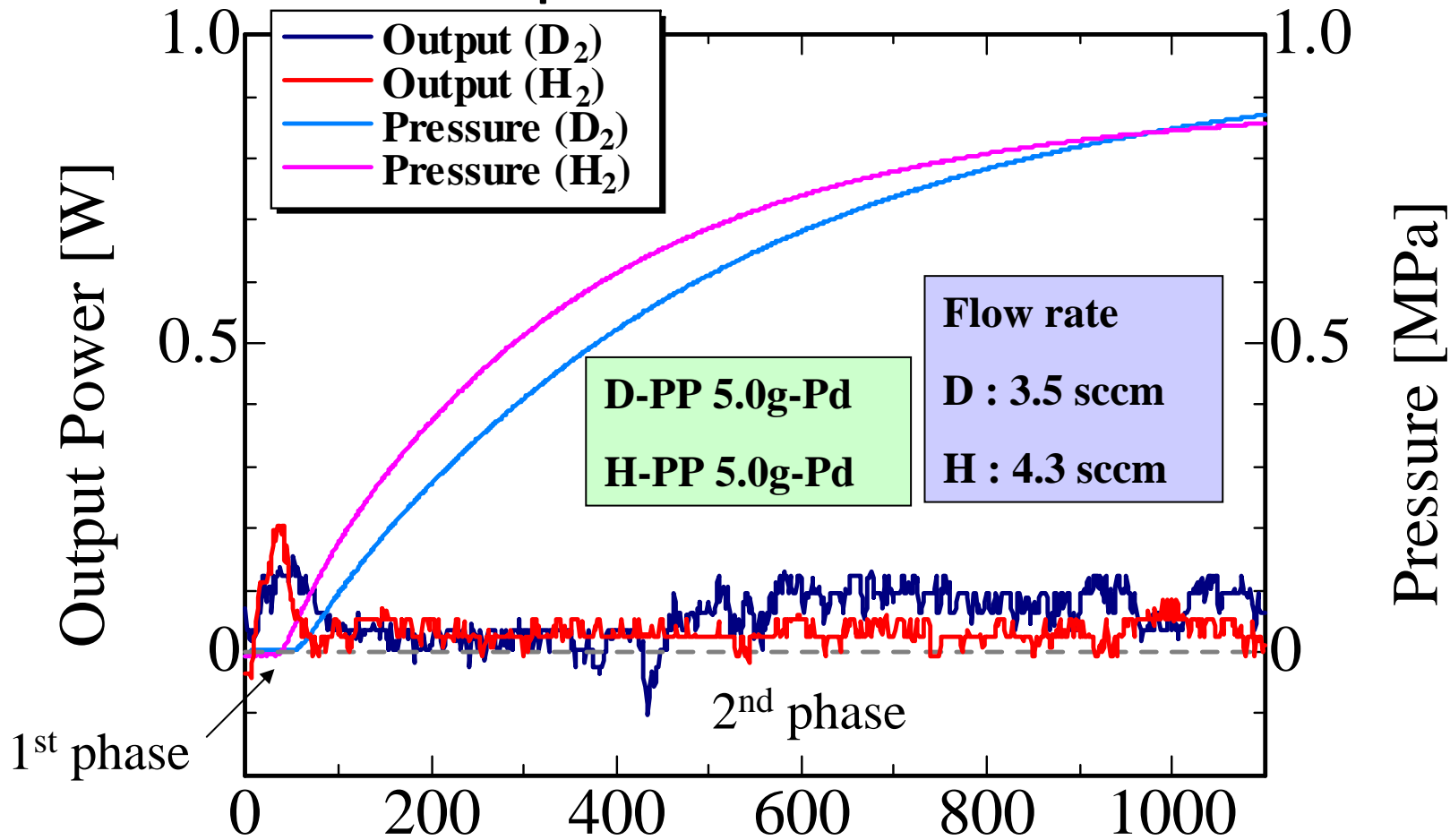
TEM Image of Santoku Pd (10.5 nm)

(By courtesy of the Nuclear Science and Engineering Institute and Particulate Systems Research Center at the University of Missouri-Columbia; Prof. R. Duncan *et al.*)

TEM images of Palladium nanoparticles from Japan (sample # 2)



$\phi 100\text{nm Pd (PP)}$

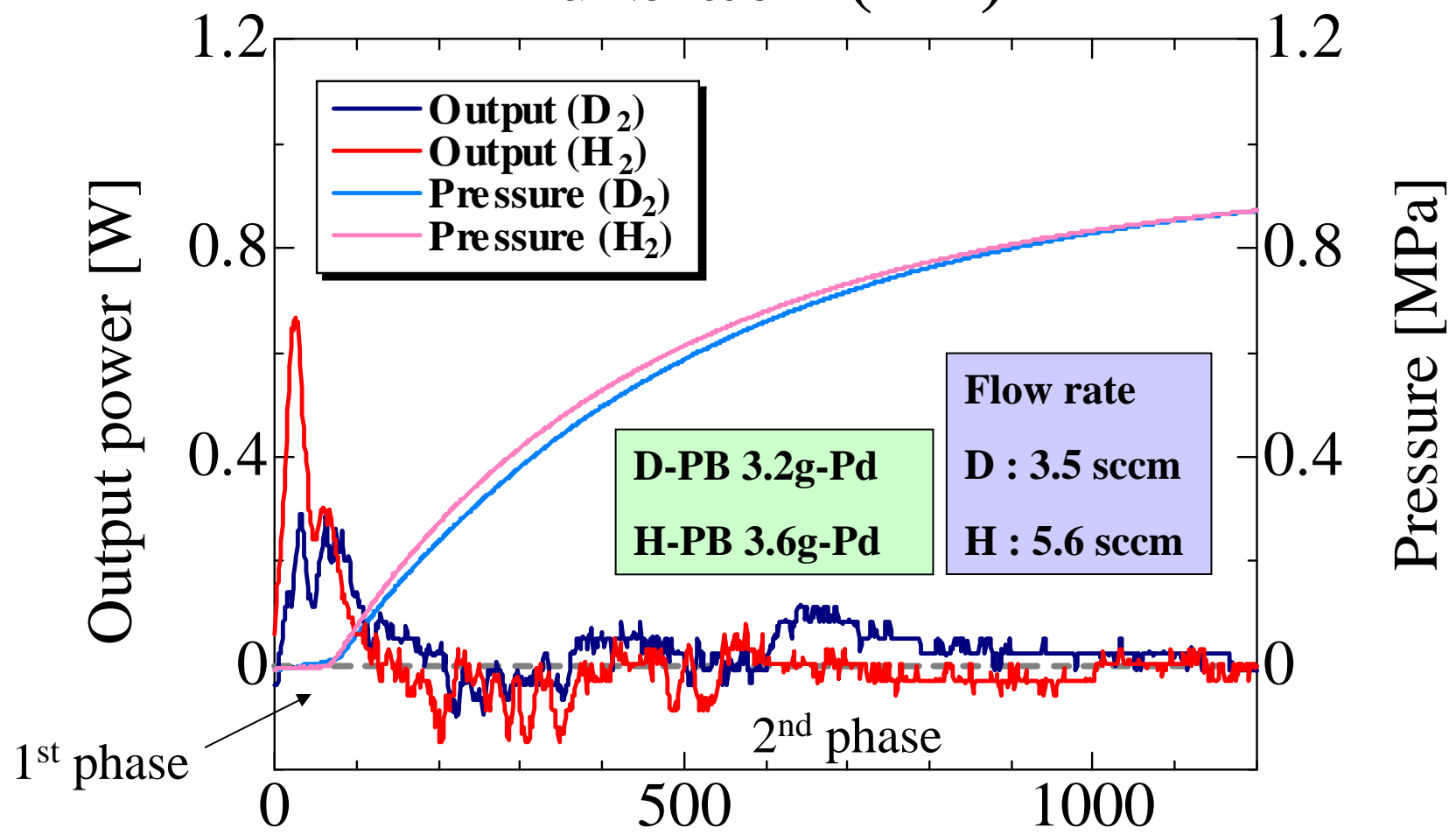


1 st phase
D ₂ : 0.10kJ/g-Pd
H ₂ : 0.08kJ/g-Pd

Time [min]
D/Pd=0.43
H/Pd=0.45

2 nd phase
D ₂ : 0.79kJ/g-Pd
H ₂ : 0.53kJ/g-Pd

Pd-black (PB)

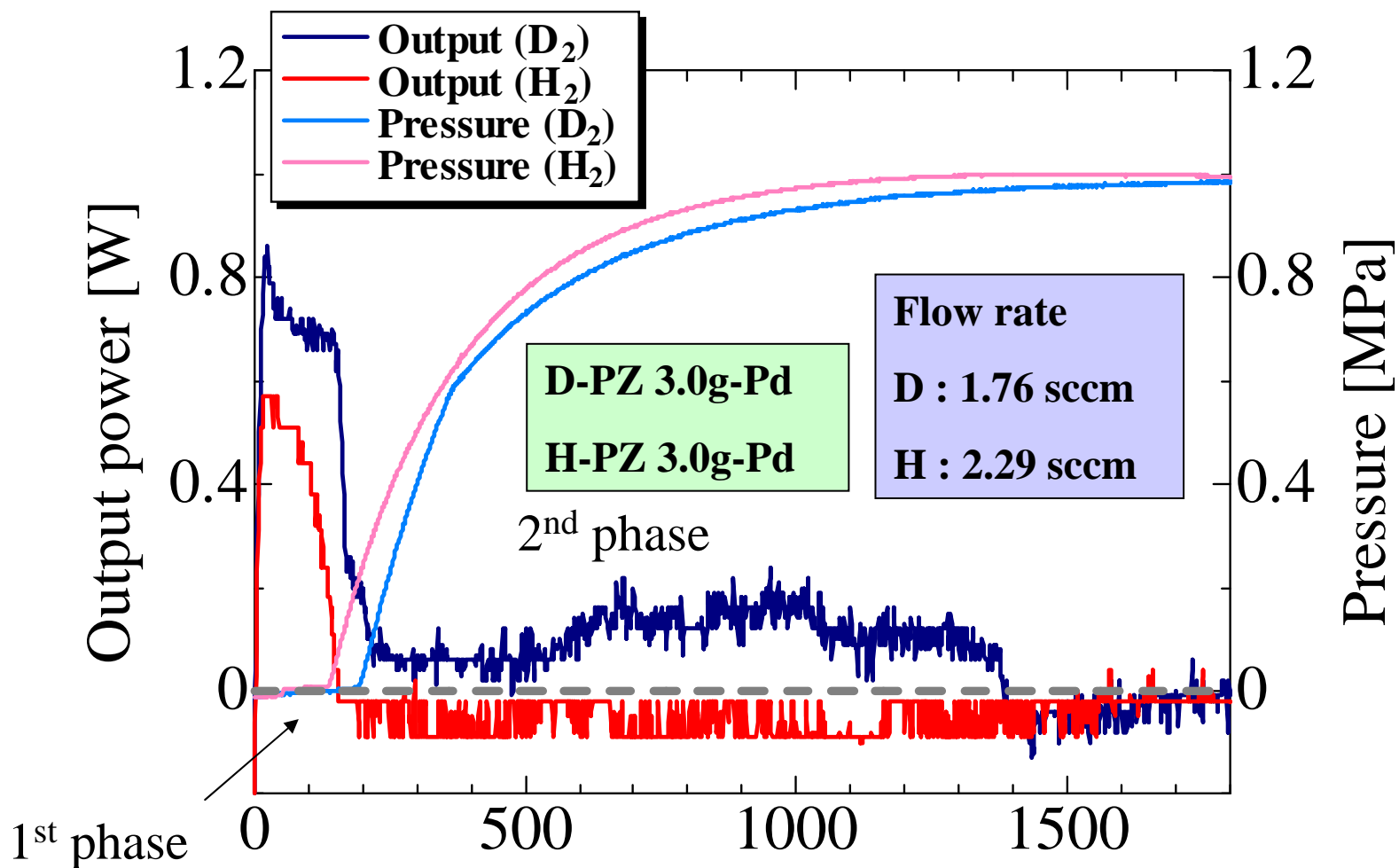


1st phase
D₂ : 0.54kJ/g-Pd
H₂ : 0.45kJ/g-Pd

Time [min]
D/Pd=0.85
H/Pd=0.78

2nd phase
D₂ : 0.65kJ/g-Pd
H₂ : -0.62kJ/g-Pd

Santoku Pd (PZ1,2#1)

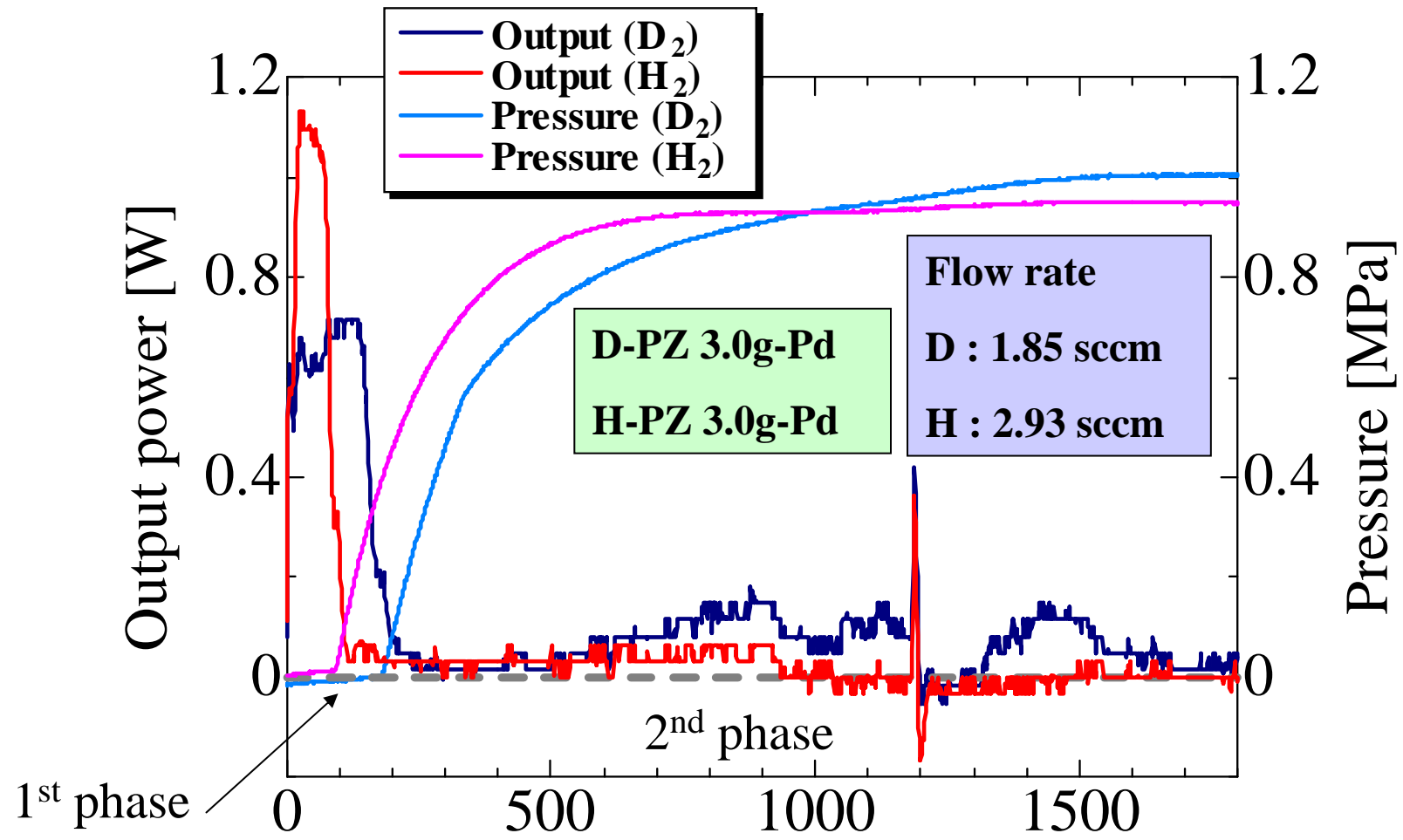


1st phase
D₂ : 1.3 kJ/g-Pd
H₂ : 1.0 kJ/g-Pd

Time [min]
D/Pd= 1.08
H/Pd= 1.00

2nd phase
D₂ : 1.9 kJ/g-Pd
H₂ : -1.3 kJ/g-Pd

Santoku Pd (PZ3,4#1)

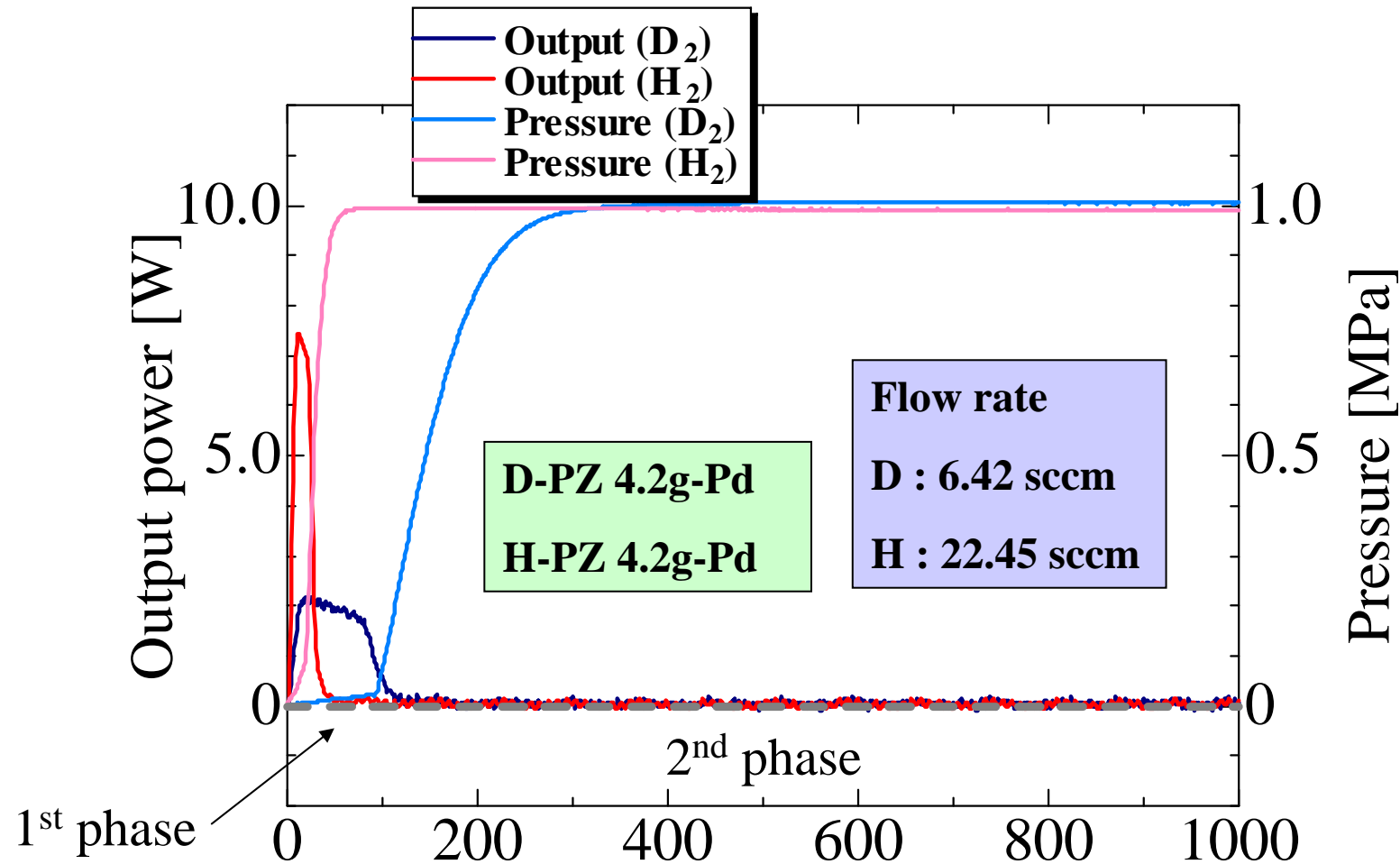


1st phase
 D₂ : 2.13kJ/g-Pd
 H₂ : 1.70kJ/g-Pd

Time [min]
 D/Pd=1.07
 H/Pd=0.86

2nd phase
 D₂ : 1.28kJ/g-Pd
 H₂ : 0.26kJ/g-Pd

Santoku Pd (PZ9,10#1)



1st phase

D_2 : 2.39kJ/g-Pd

H_2 : 2.27kJ/g-Pd

Time [min]

$D/Pd = 1.41$

$H/Pd = 1.02$

2nd phase

D_2 : 0.91 kJ/g-Pd

H_2 : 0.91kJ/g-Pd

run	weight	Gas	flow rate	Output energy[kJ]		Specific output energy[kJ/g]		D/Pd or H/Pd	E per D/H atom
	[g]		[sccm]	1st phase	2nd phase	1st phase	2nd phase		
D-PP1#1	5	D ₂	3.5	0.5±0.4	2.5±4.1	0.10±0.07	0.52±0.83	0.46	0.24
D-PP1#2	5	D ₂	4.3	0.5±0.2	4.0±4.4	0.10±0.05	0.79±0.88	0.43	0.26
H-PP2#1	5	H ₂	6.8	0.4±0.2	2.6±3.9	0.08±0.003	0.53±0.8	0.45	0.20
D-PB1#1	3.2	D ₂	3.5	1.7±0.3	8.3±4.5	0.54±0.1	2.6±1.4	0.85	0.69
H-PB2#1	3.6	H ₂	5.6	1.6±0.3	-2.2±4.6	0.45±0.08	-0.62±1.3	0.78	0.63
D-PB3#1	20	D ₂	2.9	9.3±1.1	1.1±0.5	0.47±0.06	0.058±0.023	0.78	0.66
D-PB3#2	20	D ₂	0.8	3.3±0.5	3.4±2.6	0.17±0.03	0.17±0.13	0.23	0.79
H-PB4#2	20	H ₂	1.9	3.2±0.2	14±4.6	0.16±0.01	0.68±0.23	0.22	0.80
H-PB4#3	20	H ₂	1.5	16±2.4	-4.8±8.1	0.79±0.01	-0.24±0.40	0.20	4.42
D-PB3#3	20	D ₂	1.1	14±1.7	-2.2±1.1	0.68±0.01	-1.1±0.54	0.22	3.51
D-PB3#4	20	D ₂	1.1	3.1±0.4	0.3±4.7	0.16±0.02	0.016±0.23	0.24	0.71
D-PZ1#1	10	D ₂	1.76	7.0±0.2	6.8±1.3	1.3±0.04	1.9±0.31	1.08	2.39
H-PZ2#1	10	H ₂	2.29	3.6±0.1	-5.1±1.4	1.0±0.03	-1.5±0.32	1.00	1.33
D-PZ3#1	10	D ₂	1.85	6.4±0.2	5.5±0.8	2.13±0.0	1.2±0.2	1.07	2.20
H-PZ4#1	10	H ₂	2.93	5.1±0.1	1.1±0.9	1.70±0.0	-1.3±0.2	0.86	2.18
D-PZ3#2	10	D ₂	1.66	0.17±0.03	9.89±1.48	0.03±0.070	2.3±0.35	0.29	0.13
H-PZ4#2	10	H ₂	2.79	0.58±0.05	1.68±1.46	0.17±0.011	0.39±0.34	0.31	0.59
D-PZ3#3	10	D ₂	1.69	0.29±0.04	-3.47±0.34	0.07±0.092	-0.81±0.35	0.25	0.29
H-PZ4#3	10	H ₂	2.99	0.37±0.02	0.75±0.35	0.01±0.006	0.17±0.34	0.26	0.42
D-PZ5#1	10	D ₂	2.02	7.14±0.15	1.26±1.36	2.37±0.035	0.29±0.32	1.04	2.51
H-PZ6#1	10	H ₂	6.23	7.07±0.07	-0.23±1.44	2.33±0.018	-0.05±0.33	1.41	1.82
D-PZ5#3	10	D ₂	9.93	0.54±0.025	0.23±1.51	0.18±0.008	0.08±0.50	0.25	0.74
H-PZ6#3	10	H ₂	10.69	0.92±0.025	4.18±1.51	0.31±0.008	1.39±0.50	0.30	1.10
D-PZ9#1	14	D ₂	6.42	10.23±0.10	3.81±1.51	2.44±0.024	0.91±0.36	1.41	1.87
H-PZ10#1	14	H ₂	22.55	9.56±0.034	3.82±1.51	2.28±0.008	0.91±0.36	1.02	2.46

1st phase results

run	D/Pd or H/Pd	E per D/H atom [eV]
D-PP	0.46	0.24
H-PP	0.45	0.20
D-PB	0.82±0.05	0.67±0.02
H-PB	0.78	0.63
D-PZ	1.15±0.17	2.24±0.28
H-PZ	1.07±0.24	1.95±0.49

average

- PP ; Loading ratios are bulk values, and specific heats are also bulk values.
- PB ; Loading ratios are **2-fold** of bulk values, and specific heats are **3-fold** of bulk values
- PZ ; Loading ratios are **2.5-fold** of bulk values, and specific heats are **10-fold** of bulk values

Conclusion

- **The twin system of D(H) gas loading is a useful tool.**
- **Nano-Palladium Zirconium-oxide composite generates 10-fold larger specific heat by D(H)-absorption, compared to that of bulk palladium.**
- **Nano-Palladium Zirconium-oxide composite generates excess heat in the phase-2 for D₂ gas charging.**
- **We need further to study dependence on flow rate, nano-particle size, and cell temperature.**
- **We need also study of other material samples.**
- **Analyses of ⁴He production and nuclear particle emission are expected.**