

PROGRESS TOWARDS REPLICATION

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ABSTRACT

A collaborative effort has been established formally between SRI International in Menlo Park, California, and ENEA (Frascati) in Italy. In addition to providing a framework for an International replication effort, this collaboration is intended to focus the complementary skills of the two laboratories on carefully selected problems of Pd/D studies.

Researchers are attempting to test and demonstrate the cross-laboratory replicability of gas phase Pd/D₂ excess heat, helium and tritium observations. Similar facilities are being established in both countries to allow on-line determination of heat effects correlated with ⁴He, and ultimately ³He measurements from so called “Case” experiment involving the application of modest temperatures and D₂ gas pressures to a packed bed of palladium on carbon catalyst and other finely divided metallic catalyst and conductive support materials.

A second facet of this collaboration is the joint attempt to replicate the production of tritium in “Arata/Zhang” double-structured cathodes. Massive hollow palladium electrodes were manufactured at ENEA and sealed to contain palladium black within the enclosed void. Two electrodes were operated at SRI as electrolytic cathodes in LiOD electrolyte. On experiment termination these electrodes were vented and their contents examined for ⁴He, ³He and evidence of tritium. Results of initial gas analyses are presented here. Further analyses are underway and planned.

Experiments also are underway to study the metallurgical aspects controlling the loading of D into Pd, and to measure anomalous electrical characteristics and emissions from highly loaded specimens. The results of these scientific endeavors are reported in accompanying papers [1,2].

1, INTRODUCTION

A formal collaboration has been established between SRI International and ENEA (Frascati) with the primary purpose of demonstrating experiment portability and replicability in the emerging field of Condensed Matter Nuclear Science (CMNS). Experiments are being performed to:

- Develop and understand the means to obtain, control and sustain high H/Pd and D/Pd stoichiometric ratio loading in small dimension Pd wires, foils and coatings,
- Study, thin film electrolysis cells for possible evidence of x-ray generation, tritium evolution, ³He, ⁴He and higher mass isotope production,
- Replicate observations of excess heat and ⁴He generation in Pd/C gas phase loading experiments first reported by Case [3],
- Further replicate experiments of Professors Arata and Zhang [4,5] with particular attention to replicating tritium measurements in the void space of double structured cathodes by Clarke *et al* [6].

The evidence of solid state nuclear effects has been amply demonstrated in a number of observations initially considered to be “anomalous”. The purpose of the discussion here is not to review the numerous published observations (but see, *e.g.*, [7]). Instead we simply identify the phenomena observed with high degree of certainty that are not subject to conventional (non-nuclear) interpretation.

The existence of an unaccounted for heat source in addition to known electrical and mechanical input has been demonstrated in the Pd/D system, the Ni/H(D) system and possibly others. The extent of this excess heat exceeds conventional chemical energy by at least 2-3 orders of magnitude. As such, nuclear energy production is implicated. Numerous experiments have yielded ⁴He as a near-quantitative companion of this “excess” heat. Tritium has been measured, both directly and via its decay product ³He, as a substantial but sub-quantitative product. Higher mass isotopes have been generated in some instances at isotopic ratios

differing significantly from natural. Energetic particles and radiation have been observed in the form of lattice gamma activation, X-ray production, charged particle and neutron emission [7].

There has been clear exposition of many of these phenomena at this conference, and these observations are not isolated or incoherent. Evolution of theory in the last 13 years has identified basic physical concepts and models to explain the set of observations without violating fundamental principals of physics. That is, we have exposed clearly nuclear solid state effects and developed rational bases to understand the mechanisms and rates of the underlying nuclear physics processes. At this point it is appropriate to ask: what more might we do?

The principal issue remaining is one of reproducibility, or, more specifically, lack thereof. We need to establish necessary and sufficient conditions to expose nuclear effects in condensed matter and it appears unlikely that we will establish this understanding by an empirical sifting of experimental conditions. Our collective intuition has taken us well beyond the boundaries of what initially was known, but the parameter space as yet unstudied is daunting. To establish reproducible experiments we require a fundamental understanding of the physical processes of condensed matter nuclear science.

We need a comprehensive and correct theory to assert applied physics, valid and simple diagnostics, and engineering control over key system parameters including the following:

1) Preparation of the host lattice and control of minor constituents. Some emphasis has been placed on reproducible preparation of the metallic host (e.g. Pd, Ni, Ti) for absorption of hydrogen isotopes (D, H). It has long been recognized that minor alloying or surface adsorbed elements can critically impair the host's ability to absorb or retain hydrogen [8]. More recently recognized is the role of lattice defects [9] and of synergistic minority elements [10] in establishing conditions necessary for solid state nuclear effects.

2) Establishment of high loading, chemical potential or "fuel" activity. For the loading of deuterium into bulk, macroscopic palladium it was very quickly recognized that "heroic" loadings to very high deuterium chemical potentials were necessary to stimulate nuclear processes. Early work at IMRA (Japan) [11] and SRI [12] identified a threshold loading for deuterium in palladium wires.

3) Suitable system stimulation. At ICCF5 we presented [13] an empirical function to describe excess heat production in 1 mm Pd wires loaded electrochemically with D. In addition to the expected loading threshold dependence [11, 12] and previously observed threshold dependence on current [13], a term was included to account for the flux of D through the wire surface. Although subject to previous theoretical speculation [9], this was the first clear experimental indication that excess heat, and potentially other condensed matter nuclear effects, require stimulation or perturbation from an extreme thermodynamic equilibrium or steady state. Flux of material (or electrons) across a boundary appears to provide suitable stimulation and various other means have been attempted (acoustic, magnetic, thermal flux and other) with varying results. The effect of D flux in inducing nuclear effects has been ably demonstrated [10].

2, EXPERIMENTAL

Our effort to establish reproducible experiments that can be used to prove the existence of a new phenomenon is based on: (i) fundamental theoretical understanding; (ii) clear and simple experiments; and (iii) reliable diagnostic measurements of unarguably nuclear effects.

As a suitable diagnostic we have initially focused attention on the measurement of light isotopes: ^3H , ^3He and ^4He . Considerable effort has been spent in establishing a mass spectrometric capability for this purpose. With contract support [14] we have purchased and installed (effective September 2001) a Micro-Mass 5400 noble gas mass spectrometer. This instrument has a magnetic sector mass analyzer with 90° extended geometry ion optics giving a dispersion length of 54 cm, allowing mass resolution of 0.001 dalton at low masses. Two detectors are employed to allow different ranges of sensitivity. A Faraday cup has sensitivity of ~ 3 ppb (parts per billion) for both ^3He and ^4He . The channeltron detector has sensitivity of ~ 2 ppt (parts per trillion) for ^4He and ~ 0.05 ppt for ^3He , allowing absolute resolution for ^4He of $\sim 1 \times 10^7$ atoms, and ^3He of $\sim 2.5 \times 10^5$ atoms. In general these atom sensitivities are limited by the instrument background and may be reduced by using different analysis methods.

An initial phase of experiments was performed using double structured (DS) cathodes prepared at ENEA (Frascati). A previous replication attempt at SRI resulting in copious tritium production (measured as ^3He by Clarke, Oliver *et. al.*, [6]), was performed with cathodes prepared by Professors Arata and Zhang. An attempted replication without direct benefit of their specialized knowledge and skill, while ambitious, was felt to be a better test of experiment transferability. This effort was also designed to teach us about metallurgical, machining and welding issues related to the preparation of Arata/Zhang DS cathodes:

Two cathodes, EAP1 and EPA2 were prepared with the following dimensions:

EAP1: 60 mm long x 12.5 mm diameter; mass = 78g; internal cavity; 45 mm long x 4.4 mm diameter.
Pd powder mass = 0.5 g.

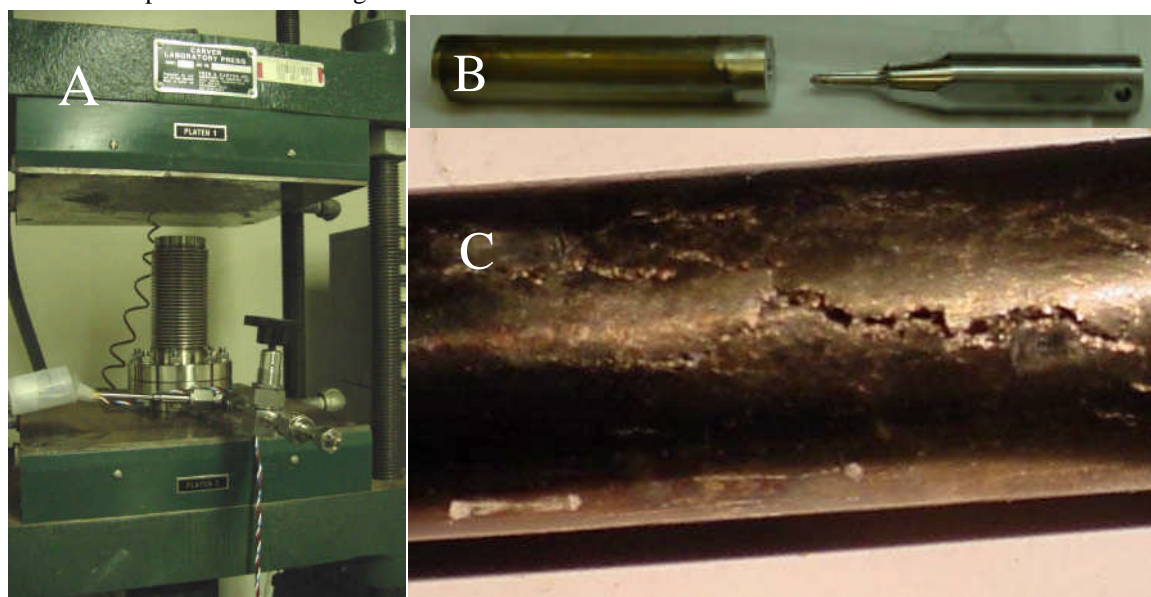


Figure 1 (A) Puncturing device in press; (B) Puncturing tip and DS cathode; (C) EAP1 crack.

EAP2: 70 mm long x 13.5 mm diameter, mass = 110g; internal cavity 45mm long x 4.4 mm diameter.
Pd powder mass = 0.6 g.

The Pd powder used to fill the internal cavity was 0.25-0.55 μ m, 99.95% (metals basis), from Alfa Aesar, Ward Hill, MA. The palladium vessels were filled with Pd powder inside an argon dry box, and then enclosed within a sealed vessel (under argon) for transfer to the electron beam welder. The Pd black filled cathodes were evacuated in the welding device at $\sim 5 \times 10^{-3}$ torr for one hour before welding closed. Ultrasound analysis revealed that the weld thickness was not less than 4mm.

The two cathodes were electrolyzed at SRI in 0.1 M LiOD roughly following Arata/Zhang current protocols [4,5]. Table 1 lists the current-time history of electrolysis. The final 35 days of anodic treatment was intended to reduce the D₂ gas pressure in the internal cavity allowing for safe handling of the electrodes.

Table 1	Current Protocol for DS Cathodes					
Current	5.0	5.5	6.0	6.5	1.0	6.0
	6.5	7.0	-0.01	-0.03	Amps	
EAP1	1031	216	241	100	504	66
	28	503	26	814	Hours	
EAP2	338	216	241	100	504	66
	28	503	26	814	Hours	

Following anodic stripping the electrodes were transferred to a device constructed at ENEA, capable of puncturing the cathode internal cavity and releasing whatever gas remained into an enclosing, evacuated chamber. By means of an attached pressure transducer it was possible to observe and quantify this gas release. Two gas samples were obtained from each electrode: the first at ambient temperature (25°C) immediately following void puncture intended to reflect the equilibrium condition of gas in the internal cavity; the second after heating of the extraction vessel to 125°C for 15 hours and for 2 hours at 250°C, to sample hydrogen isotopes absorbed in the palladium metal bulk and black.

3, RESULTS

Figure 1C shows cathode EAP1 after removal from the electrochemical cell and puncturing. The surface of the electrode exhibited considerable metallurgical effect and damage resulting from deuterium loading and deloading. At least one axial crack appears to penetrate to the internal cavity effectively preventing detection of any novel gases in the vessel space. The second cathode, EAP2, showed evidence of the surface expression of a much smaller axial crack (Figure 2). Metallurgical examination of both cathodes is planned to measure the depth of crack penetration.



Figure 2 DS Cathode EAP2 after removal from the electrochemical cell.

Table 2 presents the results of gas sampling and initial analyses. Samples were collected in the 250 cc volume of the pre-evacuated bellows puncturing device. Sample EAP1(A) thus was $\sim 1/4$ atm primarily of adsorbed D_2O and absorbed D_2 . As indicated in the last column of Table 2, the helium in this sample analyzes at the air ambient isotope ratio presumably from air-derived helium adsorbed on the high surface area Pd black.

Heating EAP1 to yield sample (B) resulted in a substantial de-sorption of D_2 from the Pd bulk and black, an associated pressure increase, and decrease in helium concentration. Because of the measurement uncertainty in [3He], this sample also displays no anomaly in helium isotope ratio.

Table 2 Preliminary Gas Determinations

Sample	Temperature ($^{\circ}C$)	Volume (cc atm)	3He (pptr) ± 0.05	4He (ppb) ± 1.0	$\frac{^3He}{^4He}$	$\frac{R}{R_A}$
EAP1 A	25	63	0.18	138	1.3×10^{-6}	1.0
EAP1 B	250	275	0.05	25	2.0×10^{-6}	1.5
EAP2 A	25	16	0.22	76	2.2×10^{-6}	2.2
EAP2 B	250	1000	< 0.05	3		

$R_A = 1.38 \times 10^{-6}$ [15]

As can be seen from the data in Table 2, sample EAP2(A) yielded a potentially more interesting result. The initial sample while only $\sim 1/4$ of the volume of EAP1(A), shows a higher concentration of 3He and a lower concentration of 4He . The helium isotope composition of this sample is significantly different from the ambient air background, suggesting that at least some of the 3He was produced during the course of the experiment.

Heating EAP2 to $250^{\circ}C$ stimulated a very significant desorption of D_2 and a pressure rise to more than 4 atm. At this dilution the uncertainties in measurement made it not possible for us to calculate the helium isotope ratio.

4, DISCUSSION

The results described above are both preliminary and incomplete; they are discussed here as the results of a "work in progress". Within the next few months we plan to make a thorough analysis of the gas contents of the four samples, and repeated analyses of 3He (and 4He) in an attempt to quantify any 3He and potentially 3H present in these samples.

A metallurgical examination is already underway at ENEA to determine the reasons for, and consequences of, electrode mechanical failure. These electrodes will also be sectioned and analyzed to determine the 3He (and 4He) content of the bulk metal phase and of the Pd black. Since the diffusion range of helium in Pd is very limited while that of hydrogen is long, careful measurement of 3He provides a very sensitive test of the past presence of tritium. The capacity to measure helium in metal samples will be established in the next phase of collaborative activity.

A new series of experiments also is planned. The $^3He/^4He$ isotope ratio measured at 2.2 times the air ambient background for sample EAP2(A) in the present series must be taken as significant and encouraging. This number is, however, substantially less than was measured [6] in our previous attempt to replicate the results of Professors Arata and Zhang. With improved measurement capability, fabrication experience gained, and expanded collaborative support, we anticipate making substantial further progress towards replication.

DISCUSSION

The authors gratefully acknowledge the substantial technical support and prowess of Mr. L. Bettinali in fabricating the DS cathodes studied in this work, and also the puncturing device used to sample internal gases.

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